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**Ensheng Dong**



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web jan 5 2016 explanation isotopes are atoms of the same element same atomic number i e same number of protons in the nucleus but different atomic masses different mass numbers due to different numbers of neutrons in the nucleus for example

we may get an isotope of chlorine with 17 protons and 18 neutrons hence atomic mass 35

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element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn't differ all that much but its mass does differ thus its density is a little

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