

Name _____

Molecular Geometries Lab Report Sheet

| Formula | Lewis Structure | No of electron domains/Atp | Electron domain geometry | VSEPR Molecular geometry | Hybridization of central atom | Polar ? |
|------------------------------|---|----------------------------|-----------------------------|-----------------------------|-------------------------------|---------|
| CO ₂ | <p>Valence 24</p> $\text{O}=\text{C}=\text{O}$ <p>2 charge clouds</p> | 0 | Linear 180° | Linear | sp | NP |
| BF ₃ | <p>24</p> $\text{F}=\text{B}-\text{F}$ | 0 | Trigonal planar 120 | | sp ² | NP |
| SnCl ₂ | <p>44</p> $\text{Cl}=\text{Sn}-\text{Cl}$ | 3 | Bent 120 | | sp ² | P |
| BF ₄ ⁻ | <p>27</p> $\left[\text{F}-\text{B}-\text{F} \right]^{-}$ | 0 | Tetrahedral 109.5 | | sp ³ | P |
| AsH ₃ | <p>4</p> $\text{H}-\text{As}-\text{H}$ | 4 | Trigonal pyramidal 109.5 | | sp ³ | P |
| H ₂ S | $\text{H}-\text{S}-\text{H}$ | 2 | Bent 109.5 | | sp ³ | NP |
| NH ₂ ⁻ | $\left[\text{H}-\text{N}-\text{H} \right]^{-}$ | 2 | Bent 109.5 | | sp ³ | P |
| PCl ₄ | <p>40</p> $\text{F}-\text{P}-\text{Cl}$ | 0 | Trigonal bipyre 120 + 90 | | sp ³ d | P |

Molecular Shapes Lab Activity Answers

Kenda Jo Lemont



Molecular Shapes Lab Activity Answers:

Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Molecular Shapes Lab Activity Answers**

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Table of Contents Molecular Shapes Lab Activity Answers

1. Understanding the eBook Molecular Shapes Lab Activity Answers
 - The Rise of Digital Reading Molecular Shapes Lab Activity Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Molecular Shapes Lab Activity Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Molecular Shapes Lab Activity Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Molecular Shapes Lab Activity Answers
 - Personalized Recommendations

- Molecular Shapes Lab Activity Answers User Reviews and Ratings
- Molecular Shapes Lab Activity Answers and Bestseller Lists
- 5. Accessing Molecular Shapes Lab Activity Answers Free and Paid eBooks
 - Molecular Shapes Lab Activity Answers Public Domain eBooks
 - Molecular Shapes Lab Activity Answers eBook Subscription Services
 - Molecular Shapes Lab Activity Answers Budget-Friendly Options
- 6. Navigating Molecular Shapes Lab Activity Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Molecular Shapes Lab Activity Answers Compatibility with Devices
 - Molecular Shapes Lab Activity Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Molecular Shapes Lab Activity Answers
 - Highlighting and Note-Taking Molecular Shapes Lab Activity Answers
 - Interactive Elements Molecular Shapes Lab Activity Answers
- 8. Staying Engaged with Molecular Shapes Lab Activity Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Molecular Shapes Lab Activity Answers
- 9. Balancing eBooks and Physical Books Molecular Shapes Lab Activity Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Molecular Shapes Lab Activity Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Molecular Shapes Lab Activity Answers
 - Setting Reading Goals Molecular Shapes Lab Activity Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Molecular Shapes Lab Activity Answers
 - Fact-Checking eBook Content of Molecular Shapes Lab Activity Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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