Question 1:

Find the radian measures corresponding to the following degree measures:
(i) 25°   (ii) −47°30′   (iii) 240°   (iv) 520°

Solution:

We know that $180° = \pi$ radian
Therefore,

(i) $\qquad 25° = \dfrac{\pi}{180} \times 25$ radian

$\qquad\qquad = \dfrac{5\pi}{36}$ radian

(ii) $\qquad -47°30′ = -\left(47\dfrac{1}{2}\right)° = -\left(\dfrac{95}{2}\right)°$

$\qquad -\left(\dfrac{95}{2}\right)° = \dfrac{\pi}{180} \times -\left(\dfrac{95}{2}\right)$ radian

$\qquad\qquad = -\dfrac{19\pi}{72}$ radian

(iii) $\qquad 245° = \dfrac{\pi}{180} \times 240$ radian

$\qquad\qquad = \dfrac{3\pi}{4}$ radian

(iv) $\qquad 520° = \dfrac{\pi}{180} \times 520$ radian

$\qquad\qquad = \dfrac{26\pi}{9}$ radian

Question 2:

Find the degree measures corresponding to the following radian measures $\left(\text{Use } \pi = \dfrac{22}{7}\right)$.

(i) $\dfrac{11}{16}$ (ii) -4 (iii) $\dfrac{5\pi}{3}$ (iv) $\dfrac{7\pi}{6}$

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web oct 29 2021 contributors and attributions learning objectives to draw lewis structures for molecules and polyatomic ions with one central atom introduction to lewis structures a lewis structure is a way to show how atoms share electrons when they form a molecule lewis structures show all of the valence electrons in an atom or molecule

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web lewis structures we also use lewis symbols to indicate the formation of covalent bonds which are shown in lewis structures drawings that describe the bonding in molecules and polyatomic ions for example when two chlorine atoms form a chlorine molecule they share one pair of electrons

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web this online quiz is intended to give you extra practice in identifying and drawing lewis dot structures as well as predicting ion formation this quiz aligns with the following ngss standard s hs ps1 1

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web jan 30 2023   a lewis structure is a very simplified representation of the valence shell electrons in a molecule it is used to show how the electrons are arranged around individual atoms in a molecule electrons are shown as dots or for bonding electrons as a line between the two atoms

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web jan 30 2023   follow these simple steps to draw lewis dot structures draw the atoms on paper and put dots around them to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

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web mar 24 2018   practice drawing lewis structures with answers and explanation the video covers the basic lewis structures for a general chemistry class the five steps a