

	A	B	C	D	E	F	G
Molecule or Polyatomic Ion	No. of Valence Electrons	Lewis Structure		Central Atom Geometry	Bond Angles	Bond Polarity	Molecular Dipole or Nonpolar
$C_2H_3Cl_2$ * 2 central atoms	24			trigonal pyramidal & tetrahedral			
HCN	10	$H-C \equiv N:$		tetrahedral			
CH_2O	12			tetrahedral			
OF_2	20			Bent			
HBr	8	$H-Br:$					
PCl_3	26			trigonal pyramidal			
NO_3^-	24			tetrahedral			

* More than one possible Lewis structure can be drawn. See questions 1, 2.

** More than one possible Lewis structure can be drawn. See question 2.

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