
Chromatography – Questions With Solutions

What is chromatography? ✓ Ans - An analytical method used to separate the substances in a mixture.

Used to identify chemicals in a mixture.

What are the two phases required in chromatography? ✓ Ans - - Stationary phase

- Mobile phase

What is the stationary phase? ✓ Ans - - Where the molecules can't move. It is a solid or a really thick liquid.

- Fixed in place

Examples of the stationary phase ✓ Ans - - Liquid= in some gas chromatography (GC) systems, where the mixture components have different attractions to each phase.

- Solid= In thin layer chromatography, where the components of the mixture are adsorbed by different amounts.

Example for TLC- thin metal sheet coated in Al_2O_3 or silica

What is the mobile phase? ✓ Ans - - Where the molecules can move. This is always a liquid or a gas.

- Moves in a definite direction

- Flows over the stationary phase

- It is a polar or non-polar liquid (solvent) that carries components of the compound being investigated.

Examples of the mobile phase ✓ Ans - - An inert gas in GC

- A liquid solvent in TLC

-Polar solvents - water or alcohol= TLC

Non-polar solvents - alkanes= TLC

Why does separation in chromatography occur? ✓ Ans - - Because each chemical in a mixture has different attractions to each phase.

Chromatography Biochemistry Questions And Answers

K Morrison



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