

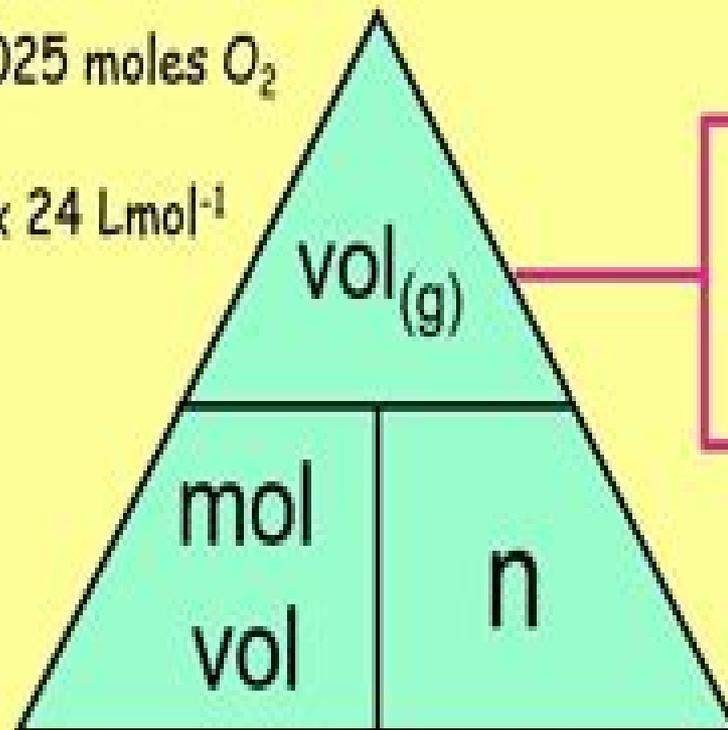
Assume molar volume is 24Lmol^{-1}

Calculate volume of 0.025 moles O_2

$$\text{Vol} = 0.025 \text{ mol} \times 24 \text{ Lmol}^{-1}$$

$$= 0.6 \text{ L}$$

$$= 600 \text{ cm}^3$$



Gas Volume

$$\text{Vol} = n \times \text{mol.vol}$$

Calculations Using Standard Molar Volume

AW Chickering



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Physical Chemistry Robert J. Silbey, Robert A. Alberty, George A. Papadantonakis, Mounqi G. Bawendi, 2022-06-15 Ever since Physical Chemistry was first published in 1913 it has remained a highly effective and relevant learning tool thanks to the efforts of physical chemists from all over the world Each new edition has benefited from their suggestions and expert advice The result of this remarkable tradition is now in your hands *Chemical Symbolism and Calculations* Stanley Wallace Morse, 1953

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Peter Atkins,Julio de Paula,Ron Friedman,2008-11-15 Beginning with quantum mechanics introducing statistical mechanics and progressing through to thermodynamics this new text for the two semester physical chemistry course features a wealth of new applications and insights as well as new Mathematical Background inter chapters to help students review key quantitative concepts This is a splendid book True to the authors philosophy as outlined in the preface it approaches physical chemistry by first developing the quantum theory of molecular electronic structure then by statistical arguments moves into thermodynamics and thence to kinetics Peter Taylor Review in Chemistry World Royal Society of Chemistry July 31 2009

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