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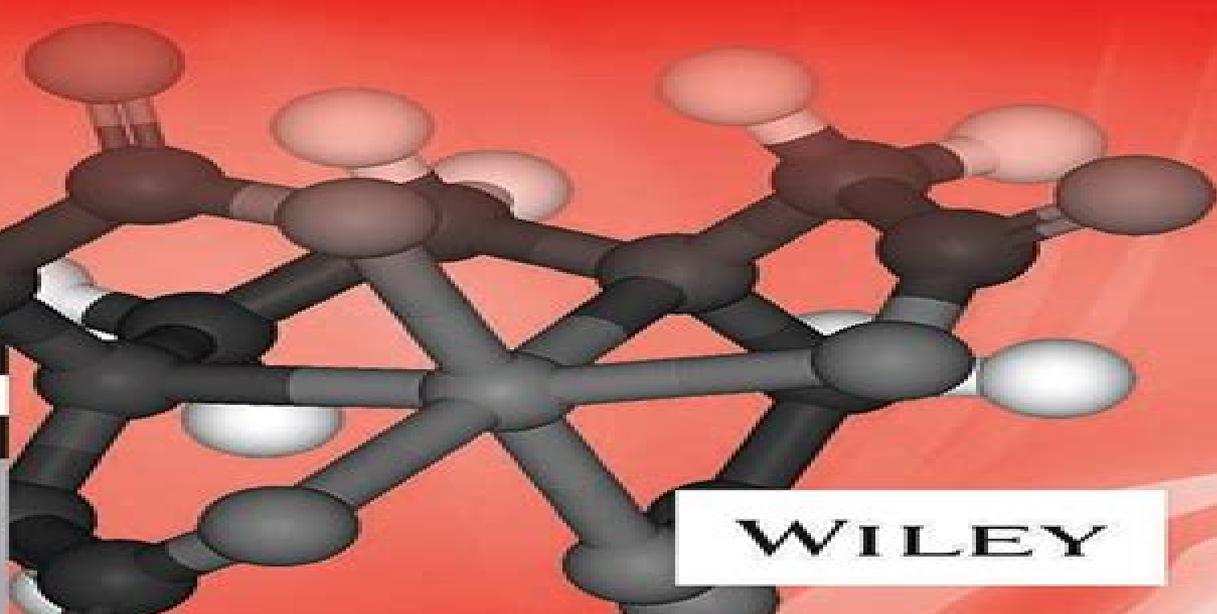
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large and very large molecules e.g. homo and heteropolymers and explains the huge variety of organic molecules. Of the silicon which is also tetravalent there are also a large number of compounds but by far not such a variety. The properties of organic substances are very much determined by their respective molecular structure. Even the properties of simple organic salts such as acetates are clearly determined by the molecular shape of the organic part. There are also many isomers i.e. compounds with the same overall composition molecular formula but different structure structural formula. In contrast molecules in inorganic chemistry usually consist of only a few atoms where the general properties of solids crystals and or ions come into play. However there are also polymers that do not contain carbon or only in subgroups e.g. the silanes. Organic synthesis strategies differ from syntheses in inorganic chemistry because organic molecules can usually be built up piece by piece. About 60% of chemists in Germany and the USA have chosen organic chemistry as their main subject.

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