

# Answer KEY

## Unit 3 Outline/ Study Guide - Electron Configuration and Periodic Trends

### Essential Skills/ State Standards:

1. Students know how to relate the position of an element in the periodic table to its quantum electron configuration and to its reactivity with other elements in the table.
2. Students know how to use the periodic table to identify metals, semimetals, nonmetals, halogens, alkali metals, alkaline earth metals, and transition metals.
3. Students know how to use the periodic table to identify the following trends:  
 ionization energy, electronegativity, and the relative sizes of ions and atoms  
 (this includes knowing the definitions of ionization energy, electronegativity, cation, and anion.)

### Study Guide - (use your notes and worksheets for a more complete review)

1. Identify the following things on the periodic table:

-metals	-alkali metals	
-metalloids (semi-metals)	-alkaline earth metals	
-transition metals	-halogens	
-nonmetals	-noble gases	
-Periods	-Groups/ families	
Identify groups with the following numbers of valence electrons: 1, 2, 3, 4, 5, 6, 7, 8		

2. Why do the elements in the groups (columns) behave so similar to each other even if they have great differences in their number of protons? **same number of valence electrons and similar chemical properties.**
3. What is the connection between the following terms: **octet rule, valence electrons, and ions?**  
**In order to achieve the octet rule, atoms gain or lose their valence electrons. They become charged ions when this occurs.**
4. a. List the elements that are exceptions to the octet rule b) Why don't these atoms achieve the octet rule?  
 a. Small atoms are happy with 2 electrons (ex: H, He, Li, Be, B)  
 b. because the 1s energy level only holds 2 electrons

# Atoms Periodic Table Review Answer Key Lbrsfs

**George Graybill**



## **Atoms Periodic Table Review Answer Key Lbrsfs:**

**Atoms, Molecules & Elements: What Are Atoms? Gr. 5-8** George Graybill, 2015-10-01 This is the chapter slice What Are Atoms from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

## Adopting the Melody of Expression: An Psychological Symphony within **Atoms Periodic Table Review Answer Key Lbrsfs**

In some sort of consumed by displays and the ceaseless chatter of immediate transmission, the melodic elegance and psychological symphony developed by the prepared word frequently diminish in to the background, eclipsed by the constant noise and disturbances that permeate our lives. However, nestled within the pages of **Atoms Periodic Table Review Answer Key Lbrsfs** a wonderful literary prize brimming with organic thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that interesting masterpiece conducts viewers on a psychological trip, well unraveling the hidden tunes and profound impact resonating within each carefully constructed phrase. Within the depths of this moving review, we shall examine the book is key harmonies, analyze their enthralling publishing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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