

## Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 2

**Q1.** What is the mass of one atom of C – 12 in grams?

- a.)  $1.992 \times 10^{-23}$  g
- b.)  $1.989 \times 10^{-23}$  g
- c.)  $1.992 \times 10^{-23}$  g
- d.)  $1.965 \times 10^{-23}$  g

**Q2.** Which of the following laws states that the total mass of the reactants is equal to the total mass of the products in a chemical reaction?

- a.) Law of multiple proportions
- b.) Law of conservation of mass
- c.) Avogadro's law
- d.) Law of constant proportions

**Q3.** The law of multiple proportions was stated by:

- a.) Dalton
- b.) Proust
- c.) Avogadro
- d.) None of the above

**Q4.** The units  $\text{kg m}^{-1} \text{s}^{-2}$  are of \_\_\_\_\_.

- a.) Force
- b.) Pressure
- c.) Acceleration
- d.) Velocity

**Q5.** The empirical formula of glucose is \_\_\_\_\_.

- a.)  $\text{CH}_2\text{O}$
- b.)  $\text{CHO}_2$
- c.)  $\text{C}_2\text{H}_2\text{O}$
- d.)  $\text{CH}_2\text{O}_2$

**Q6.** Express the following in the scientific notation:

- i.) 0.0048

# Multiple Choice Solution Chemistry 11 Questions

**Joacim Rocklöv**



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