

The Chemistry of Cookies

Chemistry Task

Get this "task" page with the accompanying file. Students complete the data table. Once completed, they will be required to write conclusions and descriptions of the experiment. They will be required to write a lab report. (1 page, 10 min)

Table 1: Empirical Molecular Formula Use the following information to calculate the mass of ingredients available from the above information. Use molecular weights of the molecules. Show your calculations.

Butter	$C_{18}H_{36}O_2$	Amount of sugar	$C_{12}H_{22}O_{11}$
Baking Soda	$NaHCO_3$	Egg	$C_5H_7NO_2$
Baking Powder	$NaHCO_3$	Vanilla extract	$C_{10}H_{12}O_2$
Brown Sugar	$C_{12}H_{22}O_{11}$	Flour	$C_6H_{10}O_5$
Wheat	$C_6H_{10}O_5$	Yeast	$C_4H_8O_2$
White Sugar	$C_{12}H_{22}O_{11}$	Pepper	$C_8H_{10}N_2O_2$
Chocolate chips	$C_{18}H_{24}O_4$	Salt	$NaCl$
Vanilla	$C_{10}H_{12}O_2$	Food coloring	$C_{10}H_{12}O_2$
Eggs	$C_5H_7NO_2$	Fat	$C_{18}H_{36}O_2$

Table 2: Yield Calculations Use the following information to determine the amount of ingredients available from the above information. Use molecular weights of the molecules. Show your calculations.

Amount of baking soda = 2.5g	Amount of sugar = 4.4 g
Amount of vanilla extract = 1.7 g	Amount of eggs = 2.8 g
Amount of flour = 10.0 g	Amount of yeast = 0.2 g
Amount of salt = 0.1 g	Amount of pepper = 0.1 g
Amount of food coloring = 0.1 g	Amount of fat = 0.1 g
Amount of chocolate chips = 0.1 g	Amount of vanilla = 0.1 g
Amount of butter = 0.1 g	Amount of flour = 0.1 g
Amount of egg = 0.1 g	Amount of salt = 0.1 g

Ingredients

- 1 cup (236.6 mL) of unsalted butter (softened)
- 1 cup (236.6 mL) of brown sugar (packed)
- 1 cup (236.6 mL) of white sugar (packed)
- 1/2 cup (118.3 mL) of vanilla extract
- 1/2 cup (118.3 mL) of egg whites (beaten)
- 1/2 cup (118.3 mL) of egg yolks (beaten)
- 1/2 cup (118.3 mL) of flour
- 1/2 cup (118.3 mL) of salt
- 1/2 cup (118.3 mL) of yeast
- 1/2 cup (118.3 mL) of pepper
- 1/2 cup (118.3 mL) of food coloring
- 1/2 cup (118.3 mL) of fat
- 1/2 cup (118.3 mL) of vanilla
- 1/2 cup (118.3 mL) of flour
- 1/2 cup (118.3 mL) of salt



Chemistry Chocolate Chip Cookies Answers

SJ Ball



Chemistry Chocolate Chip Cookies Answers:

Chemistry Chocolate Chip Cookies Answers Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Chemistry Chocolate Chip Cookies Answers**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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