

# Properties of Atoms and the Periodic Table

## Section 2 Masses of Atoms

**Preview** Section 2 of your book, using the checklist below.

- Read all section titles.
- Read all boldfaced words.
- Read all charts and graphs.
- Look at all the pictures and read their captions.
- Think about what you already know about masses of atoms.

Write three facts you learned.

1. \_\_\_\_\_
2. \_\_\_\_\_
3. \_\_\_\_\_

### Review Vocabulary

mass

**Define** mass to show its scientific meaning.

A physical property of an object that measures its gravitational attraction to other bodies.

### New Vocabulary

atomic number

Use your book or dictionary to define the following key terms.

The number of protons in an atom's nucleus.

mass number

The sum of the number of protons and neutrons in the nucleus of an atom.

isotope

Atoms of the same element that have different numbers of neutrons.

average atomic mass

The weighted average of mass of all naturally occurring isotopes of an element.

### Academic Vocabulary

define

Use a dictionary to find the scientific meaning of define.

To show or describe an element of science.

# Chapter Test Properties Of Atoms The Periodic Table Answer Key

**SJ Ball**



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