

## Single Replacement Reaction Stoichiometry

Hina K. Huynh

25 June 21

### Purpose:

To learn how to balance chemical reactions as well as determining the limiting reactant and theoretical yield. Also, be able to analyze the product of chemical reactions.

### Procedure:

During this experiment, an amount of aluminum foil will be mixed with  $\text{CuSO}_4$  or copper sulfate to see how the two chemicals react. The reaction will be observed in the production of copper metal.

### Data

#### Activity 1

#### Data Table 1

Write the balanced Chemical Equation:



|                               | <i>Answer<br/>(include the<br/>units)</i> | <i>Observations</i>                  | <i>Calculations</i>                           |
|-------------------------------|---|--------------------------------------|---|
| Volume of 1 M $\text{CuSO}_4$ | <i>100.0 mL</i>                           | <i>measured on meniscus<br/>line</i> | <i>measured with a graduated<br/>cylinder</i> |

# Solution Stoichiometry Lab Report

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