

- C. repulsion of molecules
- D. attraction of molecules
- E. surface tension of molecules.

**Answer: B**

**16. According to kinetic theory of gases, the absolute zero temperature is attained when**

- A. volume of the gas is zero
- B. pressure of the gas is zero
- C. kinetic energy of the molecules is zero
- D. specific heat of gas is zero
- E. mass is zero.

**Answer: C**

**17. Kinetic theory of gases assumes that the collisions between the molecules are**

- A. perfectly elastic
- B. perfectly inelastic
- C. partly elastic
- D. partly inelastic
- E. partly elastic and partly inelastic.

**Answer: A**

**18. The pressure of a gas in terms of its mean kinetic energy per unit volume E is equal to**

- A. E/3
- B. E/2
- C. 2E/3

**Answer: A**

**19. The**

**20. Kinetic energy of the molecules in terms of absolute temperature T is proportional to**

- A. T
- B. T<sup>2</sup>
- C. 1/T
- D. 1/T<sup>2</sup>
- E. T<sup>3/2</sup>

**Answer: A**

**21. The coefficient of expansion of a gas is**

- A. directly proportional to the pressure of the gas
- B. directly proportional to the volume of the gas
- C. inversely proportional to the pressure of the gas
- D. inversely proportional to the volume of the gas
- E. independent of the pressure and volume of the gas

**Answer: C**

**22. The coefficient of expansion of a gas is**

- A. directly proportional to the pressure of the gas
- B. directly proportional to the volume of the gas
- C. inversely proportional to the pressure of the gas

# Thermodynamics Multiple Choice Questions Answers

**LP Steffe**



**Thermodynamics Multiple Choice Questions Answers:**

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### **Table of Contents Thermodynamics Multiple Choice Questions Answers**

1. Understanding the eBook Thermodynamics Multiple Choice Questions Answers
  - The Rise of Digital Reading Thermodynamics Multiple Choice Questions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Thermodynamics Multiple Choice Questions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Thermodynamics Multiple Choice Questions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Thermodynamics Multiple Choice Questions Answers
  - Personalized Recommendations
  - Thermodynamics Multiple Choice Questions Answers User Reviews and Ratings
  - Thermodynamics Multiple Choice Questions Answers and Bestseller Lists

5. Accessing Thermodynamics Multiple Choice Questions Answers Free and Paid eBooks
  - Thermodynamics Multiple Choice Questions Answers Public Domain eBooks
  - Thermodynamics Multiple Choice Questions Answers eBook Subscription Services
  - Thermodynamics Multiple Choice Questions Answers Budget-Friendly Options
6. Navigating Thermodynamics Multiple Choice Questions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Thermodynamics Multiple Choice Questions Answers Compatibility with Devices
  - Thermodynamics Multiple Choice Questions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Thermodynamics Multiple Choice Questions Answers
  - Highlighting and Note-Taking Thermodynamics Multiple Choice Questions Answers
  - Interactive Elements Thermodynamics Multiple Choice Questions Answers
8. Staying Engaged with Thermodynamics Multiple Choice Questions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Thermodynamics Multiple Choice Questions Answers
9. Balancing eBooks and Physical Books Thermodynamics Multiple Choice Questions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Thermodynamics Multiple Choice Questions Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Thermodynamics Multiple Choice Questions Answers
  - Setting Reading Goals Thermodynamics Multiple Choice Questions Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Thermodynamics Multiple Choice Questions Answers
  - Fact-Checking eBook Content of Thermodynamics Multiple Choice Questions Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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