

## MCQ Module - Understanding Chemical Equilibrium Practice

### Instructions:

Answer the three multiple-choice questions below. You will receive an instant feedback message when you have completed the questions and your answers have been marked.

### Question 1/3/3

1. Which does NOT describe a closed system?
- a) A thermodynamic system that can exchange energy with its surroundings
  - b) A thermodynamic system that can exchange matter with its surroundings
  - c) A thermodynamic system that can exchange both energy and matter with its surroundings
  - d) A thermodynamic system that can exchange energy with its surroundings
2. Which description of a substance at equilibrium with its surroundings is NOT correct?
- a) The substance has a constant temperature
  - b) The substance has a constant pressure
  - c) The substance has a constant volume
  - d) The substance has a constant mass
3. Which has the highest boiling point for a pure substance?
- a) 100
  - b) 1000
  - c) 10000
  - d) 100000
4. Which does NOT describe a closed system at equilibrium with its surroundings?
- a) The substance has a constant volume
  - b) The substance has a constant mass
  - c) The substance has a constant pressure
  - d) The substance has a constant temperature
5. Which is NOT a characteristic of a closed system at equilibrium with its surroundings?
- a) The substance has a constant mass
  - b) The substance has a constant volume
  - c) The substance has a constant pressure
  - d) The substance has a constant temperature

# Basic Ict Multiple Choice Questions And Answers

**James Dixon**



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