

Introduction

1. The following are the main points of the article (using the same numbering):

1.1. 1.1

The first point is that the article is a review of the literature on the topic of business ethics and corporate social responsibility. It is not a theoretical article, but rather a practical one, aimed at providing a comprehensive overview of the current state of the field.

2. There is a significant gap in the current research on business ethics.

2.1. 2.1

There is a significant gap in the current research on business ethics. The existing literature is largely descriptive and lacks a theoretical framework that can guide future research.

3. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

3.1. 3.1

The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

4. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

4.1. 4.1

The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

5. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

5.1. 5.1

The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

6. There is a significant gap in the current research on business ethics. The existing literature is largely descriptive and lacks a theoretical framework that can guide future research.

6.1. 6.1

There is a significant gap in the current research on business ethics. The existing literature is largely descriptive and lacks a theoretical framework that can guide future research. The article is organized as follows: Section 2 discusses the current state of the field, Section 3 discusses the theoretical framework, and Section 4 discusses the research methodology.

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balancing redox reactions in acidic solution problems 1 10 problem 1 $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} \rightarrow \text{Cr}^{3+} + \text{Fe}^{3+}$ solution 1 balanced half reactions $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^-$ equalize the electrons $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $6\text{Fe}^{2+} \rightarrow 6\text{Fe}^{3+} + 6\text{e}^-$ multiplied by a factor of 6

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web balance each half reaction in basic solution 4 5 6 7 $\text{Cr}_2\text{O}_7^{2-} + \text{NO}_2 \rightarrow \text{Cr}^{3+} + \text{NO}_3^-$ balance each redox reaction in acid solution using the half reaction method 8 $\text{H}_2\text{O}_2 + \text{Cr}_2\text{O}_7^{2-} \rightarrow \text{TeO}_3 + \text{N}_2\text{O}_4 + \text{O}_2$ $\text{Cr}_3 + \text{Te} \rightarrow \text{TeO}_4$ 11 $\text{PbO}_2 + \text{IO}_3^- \rightarrow \text{IO}_2 + \text{Pb}^{2+}$ 12 $\text{AsO}_3 + \text{IO}_3^- \rightarrow \text{IO}_2 + \text{H}_2\text{AsO}_4$

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web practice exercises balance the following equations of redox reactions assign oxidation numbers to all elements in the reaction separate the redox reaction into two half reactions balance the atoms in each half reaction add the two half reactions together and cancel out common terms

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H_2O H^+ and e^-

in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively c 6 h 8 o 6 2h 2no 2 c 6 h 6 o 6 2h 2 o 2no choose 1 answer c 6 h 8 o 6 and no 2 a c 6 h 8 o 6 and no 2 no 2 and c 6 h 8 o 6 b no 2 and c 6 h 8 o 6 h and no 2 c h and