

- 4 Chlorine has two isotopes, ${}^{35}_{17}\text{Cl}$ and ${}^{37}_{17}\text{Cl}$.
- (a) What do you understand by the term *isotope*? [1]
- (b) Write down the number of neutrons in each of these isotopes of chlorine. [2]
- (c) Draw and label a diagram of an atom of chlorine to show:
- the nucleus
 - the electron shells, with the correct number of electrons in each shell. [3]
- (d) A chloride ion has eight electrons in its outer shell. What is the importance of this electronic configuration? [1]

(Paper 3)

Chapter 2 Section Review Questions

Lei Huang



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Table of Contents Chapter 2 Section Review Questions

1. Understanding the eBook Chapter 2 Section Review Questions
 - The Rise of Digital Reading Chapter 2 Section Review Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 2 Section Review Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 2 Section Review Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 2 Section Review Questions

- Personalized Recommendations
 - Chapter 2 Section Review Questions User Reviews and Ratings
 - Chapter 2 Section Review Questions and Bestseller Lists
5. Accessing Chapter 2 Section Review Questions Free and Paid eBooks
 - Chapter 2 Section Review Questions Public Domain eBooks
 - Chapter 2 Section Review Questions eBook Subscription Services
 - Chapter 2 Section Review Questions Budget-Friendly Options
 6. Navigating Chapter 2 Section Review Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 2 Section Review Questions Compatibility with Devices
 - Chapter 2 Section Review Questions Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 2 Section Review Questions
 - Highlighting and Note-Taking Chapter 2 Section Review Questions
 - Interactive Elements Chapter 2 Section Review Questions
 8. Staying Engaged with Chapter 2 Section Review Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 2 Section Review Questions
 9. Balancing eBooks and Physical Books Chapter 2 Section Review Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 2 Section Review Questions
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chapter 2 Section Review Questions
 - Setting Reading Goals Chapter 2 Section Review Questions
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chapter 2 Section Review Questions

- Fact-Checking eBook Content of Chapter 2 Section Review Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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