

# Chemistry

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(Chapter 11) (Alcohols Phenols and Ethers) (Exercises)

(Class 12)



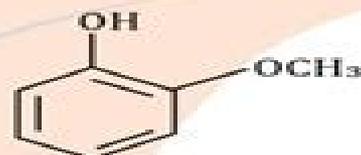
## Question 11.15:

Explain why is *ortho* nitrophenol more acidic than *ortho* methoxyphenol?

Answer 11.15:



*o* - Nitrophenol



*o* - Methoxyphenol

The nitro-group is an electron-withdrawing group. The presence of this group in the *ortho* position decreases the electron density in the O-H bond. As a result, it is easier to lose a proton. Also, the *o*-nitrophenoxide ion formed after the loss of protons is stabilized by resonance. Hence, *ortho* nitrophenol is a stronger acid.

On the other hand, methoxy group is an electron-releasing group. Thus, it increases the electron density in the O-H bond and hence, the proton cannot be given out easily.

For this reason, *ortho*-nitrophenol is more acidic than *ortho*-methoxyphenol.

## Question 11.16:

Explain how does the -OH group attached to a carbon of benzene ring activate it towards electrophilic substitution?

Answer 11.16:

The -OH group is an electron-donating group. Thus, it increases the electron density in the benzene ring as shown in the given resonance structure of phenol.



As a result, the benzene ring is activated towards electrophilic substitution.

## Question 11.17:

Give equations of the following reactions:

- (i) Oxidation of propan -1-ol with alkaline  $\text{KMnO}_4$  solution.
- (ii) Bromine in  $\text{CS}_2$  with phenol.
- (iii) Dilute  $\text{HNO}_3$  with phenol.
- (iv) Treating phenol with chloroform in presence of aqueous  $\text{NaOH}$ .

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# Nelson Chemistry 12 Chapter 5 Solutions

**V.M. Malhotra**



## **Nelson Chemistry 12 Chapter 5 Solutions:**

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*Recent Advances in Nanoscience and Technology* Sunil Kumar Bajpai, Murali Mohan Yallapu, 2009 The present Ebook deals with various strategies that have frequently been followed to fabricate nanostructures of required size and shape and with required functionalities to enable them to be used in a wide spectrum of industrial, biomedical, and technological applications. **Natural and Enhanced Attenuation of Contaminants in Soils, Second Edition** Raymond N. Yong, Catherine N. Mulligan, 2019-04-11 Natural attenuation has become an effective and low-cost alternative to more expensive engineered remediation. This new edition updates the principles and fundamentals of natural attenuation of contaminants with a broader view of the field. It includes new methods for evaluating natural attenuation mechanisms and microbial activity at the lab and field scales. Case studies, actual treatments, and protocols, theoretical processes, case studies, numerical models, and legal aspects in the natural attenuation of organic and inorganic contaminants are examined. Challenges and future directions for the implementation of natural attenuation and enhanced remediation techniques are also considered. **Rearrangement of Organosilylhydrazine Anions and Investigation of Organosilyl Hydrazyl Radicals** Blake Robert Bichlmeier, 1973 [The Analytical Chemistry of Potassium Cyanate with Particular Emphasis on Its Determination in Biological Materials](#) Nile Nelson Frawley, 1977 [Condensed Silica Fume in Concrete](#) V.M. Malhotra, 2018-01-18 As the first of its kind, this book presents a balanced view of the effect of condensed silica fume on the physical, chemical, mechanical, and durability aspects with respect

to cement paste mortar and concrete It discusses the nature and types of condensed silica fume physical characteristics product variation and problems involved in its handling and transportation **Children's Books in Print, 2007** ,2006

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**Table of Contents Nelson Chemistry 12 Chapter 5 Solutions**

1. Understanding the eBook Nelson Chemistry 12 Chapter 5 Solutions
  - The Rise of Digital Reading Nelson Chemistry 12 Chapter 5 Solutions
  - Advantages of eBooks Over Traditional Books
2. Identifying Nelson Chemistry 12 Chapter 5 Solutions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Nelson Chemistry 12 Chapter 5 Solutions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Nelson Chemistry 12 Chapter 5 Solutions
  - Personalized Recommendations
  - Nelson Chemistry 12 Chapter 5 Solutions User Reviews and Ratings
  - Nelson Chemistry 12 Chapter 5 Solutions and Bestseller Lists
5. Accessing Nelson Chemistry 12 Chapter 5 Solutions Free and Paid eBooks
  - Nelson Chemistry 12 Chapter 5 Solutions Public Domain eBooks
  - Nelson Chemistry 12 Chapter 5 Solutions eBook Subscription Services
  - Nelson Chemistry 12 Chapter 5 Solutions Budget-Friendly Options
6. Navigating Nelson Chemistry 12 Chapter 5 Solutions eBook Formats
  - ePub, PDF, MOBI, and More
  - Nelson Chemistry 12 Chapter 5 Solutions Compatibility with Devices
  - Nelson Chemistry 12 Chapter 5 Solutions Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Nelson Chemistry 12 Chapter 5 Solutions
  - Highlighting and Note-Taking Nelson Chemistry 12 Chapter 5 Solutions
  - Interactive Elements Nelson Chemistry 12 Chapter 5 Solutions
8. Staying Engaged with Nelson Chemistry 12 Chapter 5 Solutions

- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Nelson Chemistry 12 Chapter 5 Solutions
9. Balancing eBooks and Physical Books Nelson Chemistry 12 Chapter 5 Solutions
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Nelson Chemistry 12 Chapter 5 Solutions
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Nelson Chemistry 12 Chapter 5 Solutions
- Setting Reading Goals Nelson Chemistry 12 Chapter 5 Solutions
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Nelson Chemistry 12 Chapter 5 Solutions
- Fact-Checking eBook Content of Nelson Chemistry 12 Chapter 5 Solutions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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