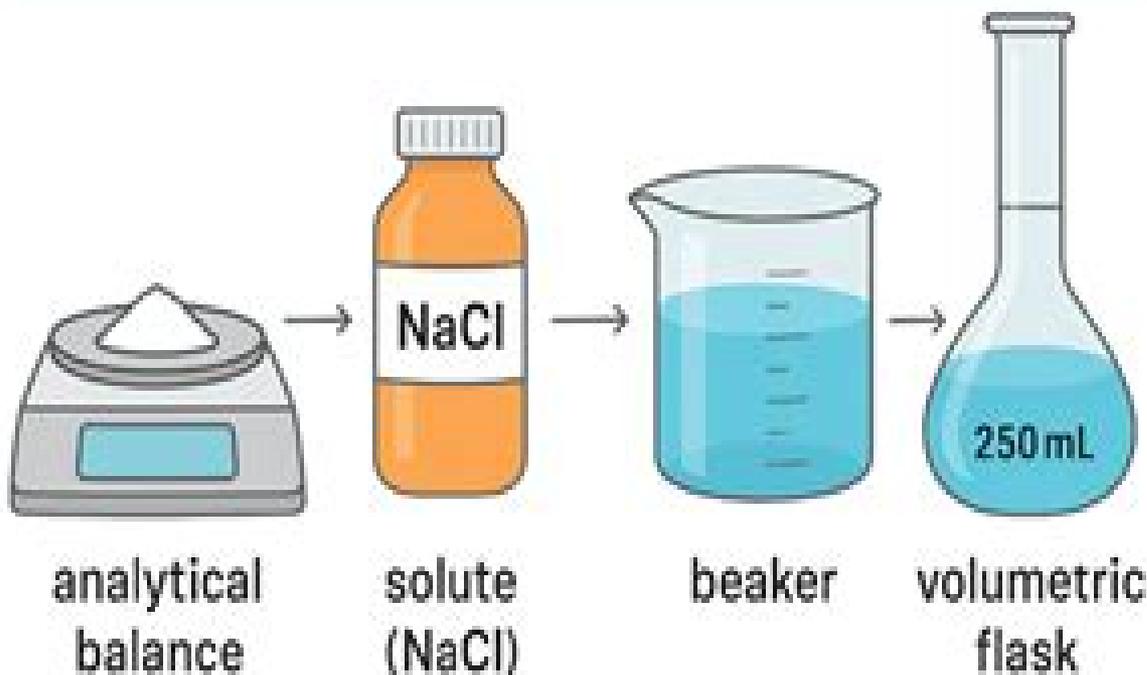


MOLARITY

Molarity (M) is the number of moles of solute per liter of solution.

$$M = \frac{n \text{ (moles of solute)}}{V \text{ (volume of solution)}}$$



Molarity Of A Solution Definition

Daniel L. Reger



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Textbook of Chemistry Albert Lawrence Elder, 1948 **Basic Chemistry** Steven S. Zumdahl, 1999 For a full description see catalog entry for Zumdahl Introductory Chemistry A Foundation 4 e [Study Guide and Glossary to Accompany Chemical Principles](#) Daniel L. Reger, Edward E. Mercer, Robert S. Boikess, 1978 **Introductory Chemistry** Steven S. Zumdahl, 1996 **Chemistry** James C. Hill, 2003 This book assists students through the text material with chapter overviews learning objectives review of key terms cumulative chapter review quizzes and self tests Included are answers to

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