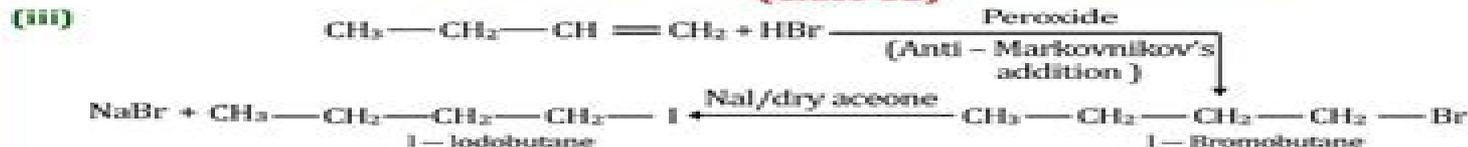


# Chemistry

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(Chapter 6) (Haloalkanes and Haloarenes) (Exercises)

(Class 12)



### Question 6.8:

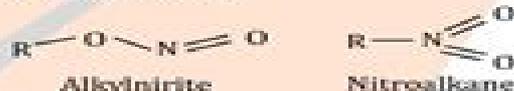
What are ambident nucleophiles? Explain with an example.

### Answer 6.8:

Ambident nucleophiles are nucleophiles having two nucleophilic sites. Thus, ambident nucleophiles have two sites through which they can attack. For example, nitrite ion is an ambident nucleophile.



Nitrite ion can attack through oxygen resulting in the formation of alkyl nitrites. Also, it can attack through nitrogen resulting in the formation of nitroalkanes.



### Question 6.9:

Which compound in each of the following pairs will react faster in  $\text{S}_{\text{N}}2$  reaction with  $\text{OH}^-$ ?

(i)  $\text{CH}_3\text{Br}$  or  $\text{CH}_3\text{I}$

(ii)  $(\text{CH}_3)_3\text{CCl}$  or  $\text{CH}_3\text{Cl}$

### Answer 6.9:

(i) In the  $\text{S}_{\text{N}}2$  mechanism, the reactivity of halides for the same alkyl group increases in the order. This happens because as the size increases, the halide ion becomes a better leaving group.



Therefore,  $\text{CH}_3\text{I}$  will react faster than  $\text{CH}_3\text{Br}$  in  $\text{S}_{\text{N}}2$  reactions with  $\text{OH}^-$ .

(ii)



The  $\text{S}_{\text{N}}2$  mechanism involves the attack of the nucleophile at the atom bearing the leaving group. But, in case of  $(\text{CH}_3)_3\text{CCl}$ , the attack of the nucleophile at the carbon atom is hindered because of the presence of bulky substituents on that carbon atom bearing the leaving group. On the other hand, there are no bulky substituents on the carbon atom bearing the leaving group in  $\text{CH}_3\text{Cl}$ . Hence,  $\text{CH}_3\text{Cl}$  reacts faster than  $(\text{CH}_3)_3\text{CCl}$  in  $\text{S}_{\text{N}}2$  reaction with  $\text{OH}^-$ .

### Question 6.10:

Predict all the alkenes that would be formed by dehydrohalogenation of the following halides with sodium ethoxide in ethanol and identify the major alkene:

(i) 1-Bromo-1-methylcyclohexane

(ii) 2-Chloro-2-methylbutane

(iii) 2,2,3-Trimethyl-3-bromopentane.

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1. Understanding the eBook Modern Chemistry Chapter 6 Harcobo
- The Rise of Digital Reading Modern Chemistry Chapter 6 Harcobo
- Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 6 Harcobo
- Exploring Different Genres
- Considering Fiction vs. Non-Fiction
- Determining Your Reading Goals
3. Choosing the Right eBook Platform
- Popular eBook Platforms
- Features to Look for in an Modern Chemistry Chapter 6 Harcobo
- User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 6 Harcobo
- Personalized Recommendations
- Modern Chemistry Chapter 6 Harcobo User Reviews and Ratings

- Modern Chemistry Chapter 6 Harcobo and Bestseller Lists
- 5. Accessing Modern Chemistry Chapter 6 Harcobo Free and Paid eBooks
  - Modern Chemistry Chapter 6 Harcobo Public Domain eBooks
  - Modern Chemistry Chapter 6 Harcobo eBook Subscription Services
  - Modern Chemistry Chapter 6 Harcobo Budget-Friendly Options
- 6. Navigating Modern Chemistry Chapter 6 Harcobo eBook Formats
  - ePub, PDF, MOBI, and More
  - Modern Chemistry Chapter 6 Harcobo Compatibility with Devices
  - Modern Chemistry Chapter 6 Harcobo Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 6 Harcobo
  - Highlighting and Note-Taking Modern Chemistry Chapter 6 Harcobo
  - Interactive Elements Modern Chemistry Chapter 6 Harcobo
- 8. Staying Engaged with Modern Chemistry Chapter 6 Harcobo
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modern Chemistry Chapter 6 Harcobo
- 9. Balancing eBooks and Physical Books Modern Chemistry Chapter 6 Harcobo
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modern Chemistry Chapter 6 Harcobo
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Chapter 6 Harcobo
  - Setting Reading Goals Modern Chemistry Chapter 6 Harcobo
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Chapter 6 Harcobo
  - Fact-Checking eBook Content of Modern Chemistry Chapter 6 Harcobo
  - Distinguishing Credible Sources

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  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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