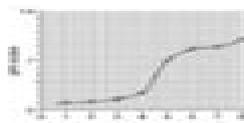


4. State the colour of universal indicator in solutions of
- an acid \_\_\_\_\_ (1 mark)
  - an alkali \_\_\_\_\_ (1 mark)
5. a Do acids taste sweet or sour? \_\_\_\_\_ (1 mark)
- b Which is dangerous, acid, alkali or they can both be dangerous? \_\_\_\_\_ (1 mark)

6. Aaf' wanted to investigate the pH changes when an acid reacts with an alkali.

He put 5 cm<sup>3</sup> of the acid in a tube.  
He added the alkali 1 cm<sup>3</sup> at a time.  
Each time he added the alkali, he measured the pH of the solution.  
His results are shown on the graph.



- a Suggest a way in which Aaf' could have measured the pH of the solution. \_\_\_\_\_ (1 mark)
- b State the volume of alkali Aaf' needed to make the solution neutral. \_\_\_\_\_ (1 mark)
- c Explain what the graph shows. \_\_\_\_\_ (2 marks)

7. A substance has a pH of 12.  
State the hazard symbol that should be on a bottle. \_\_\_\_\_ (1 mark)
- Is the substance an acid or an alkali? \_\_\_\_\_ (1 mark)

8. An indicator called bromothymol blue is yellow in an acid and blue in an alkali.  
Suggest the colour of bromothymol blue in a neutral solution. \_\_\_\_\_ (1 mark)

9. The word equation for a reaction is  
nitric acid + potassium hydroxide → potassium nitrate + water  
a State the names of the two reactants in the word equation. \_\_\_\_\_ (2 marks)
- b State the name of the salt in the word equation. \_\_\_\_\_ (1 mark)

10. Hydrochloric acid reacts with calcium oxide to make calcium chloride and water.  
Write the word equation for this reaction. \_\_\_\_\_ (2 marks)

11. Elen investigated the temperature rise when different acids were added to sodium hydroxide solution.

She measured 10 cm<sup>3</sup> of sodium hydroxide solution, and noted the temperature.  
She added 10 cm<sup>3</sup> of an acid to the sodium hydroxide solution and noted the new temperature.  
She calculated the temperature rise.  
She repeated the experiment with the same volume of the sodium hydroxide solution but added different acids to it.  
Her results are in the table.

Acid added to sodium hydroxide solution	Temperature rise (°C)
nitric acid	5.0
hydrochloric acid	13.0
nitric acid	13.0
sulphuric acid	8.0

6. Answer the following questions using one of the substances in the box below.
- |            |              |           |       |               |
|------------|--------------|-----------|-------|---------------|
| pure water | citrus juice | indicator | sugar | soapy cleaner |
|------------|--------------|-----------|-------|---------------|
- a Which substance contains an acid? \_\_\_\_\_ (1 mark)
- b Which substance contains an alkali? \_\_\_\_\_ (1 mark)
- c Which substance changes colour to show an acid or an alkali? \_\_\_\_\_ (1 mark)

7. State what each of these hazard symbols means.

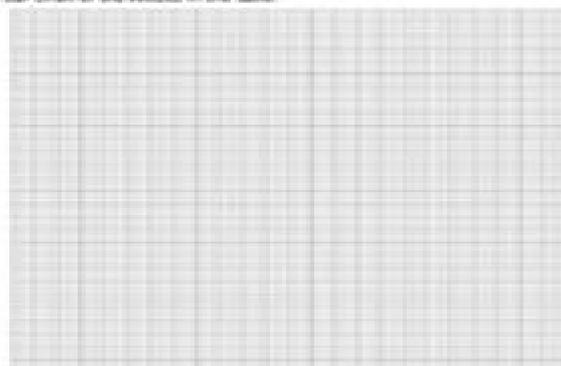
a  \_\_\_\_\_ (1 mark)

b  \_\_\_\_\_ (1 mark)

c  \_\_\_\_\_ (1 mark)

8. A bottle of acid is labelled with this hazard symbol.  
  
State two safety precautions you would take before using that acid in an experiment.  
Precaution 1 \_\_\_\_\_ (1 mark)

- a State which variable has been changed in this experiment (the independent variable). \_\_\_\_\_ (1 mark)
- b State which variable has been measured (the dependent variable). \_\_\_\_\_ (1 mark)
- c State the word that describes the reaction between an acid and an alkali. \_\_\_\_\_ (1 mark)
- d Plot a bar chart of the results in the table. \_\_\_\_\_ (2 marks)



e Write a title for the chart. \_\_\_\_\_ (1 mark)

# Exploring Science 7i End Of Unit Test Pdfsdocuments2

**R Sandford**



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### **Table of Contents Exploring Science 7i End Of Unit Test Pdfsdocuments2**

1. Understanding the eBook Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - The Rise of Digital Reading Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Advantages of eBooks Over Traditional Books
2. Identifying Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - User-Friendly Interface
4. Exploring eBook Recommendations from Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Personalized Recommendations
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 User Reviews and Ratings
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 and Bestseller Lists

5. Accessing Exploring Science 7i End Of Unit Test Pdfsdocuments2 Free and Paid eBooks
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 Public Domain eBooks
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 eBook Subscription Services
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 Budget-Friendly Options
6. Navigating Exploring Science 7i End Of Unit Test Pdfsdocuments2 eBook Formats
  - ePub, PDF, MOBI, and More
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 Compatibility with Devices
  - Exploring Science 7i End Of Unit Test Pdfsdocuments2 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Highlighting and Note-Taking Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Interactive Elements Exploring Science 7i End Of Unit Test Pdfsdocuments2
8. Staying Engaged with Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Exploring Science 7i End Of Unit Test Pdfsdocuments2
9. Balancing eBooks and Physical Books Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Exploring Science 7i End Of Unit Test Pdfsdocuments2
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Setting Reading Goals Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Fact-Checking eBook Content of Exploring Science 7i End Of Unit Test Pdfsdocuments2
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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