

### balance nuclear reaction equations

aluminum-26 undergoes positron emission  
 potassium-40 undergoes beta decay (electron)  
 uranium-238 undergoes alpha decay



### electron configurations

Write the electron configuration for C.  
 Write the electron configuration for Cr.  
 Write the shorthand notation for Bi.



### NOMENCLATURE (naming compounds, writing formulas)

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride  $\text{SF}_6$
- b. Lithium nitride  $\text{Li}_3\text{N}$
- c. Chromium (III) carbonate  $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride  $\text{SnCl}_2$
- e. Ammonium acetate  $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$  (or  $\text{CH}_3\text{COONH}_4$ )
- f. Mercury (I) chloride  $\text{Hg}_2\text{Cl}_2$
- g. Potassium bromate  $\text{KBrO}_3$
- h. Hydrosulfuric acid  $\text{H}_2\text{S(aq)}$
- i. Chloric acid  $\text{HClO}_3$
- j. Sulfurous acid  $\text{H}_2\text{SO}_3$

Name each of the following compounds:

- a.  $\text{CuSO}_4$  Copper (II) sulfate
- b.  $\text{AlF}_3$  aluminum fluoride
- c.  $\text{HI}$  hydrogen iodide
- d.  $\text{NO}$  nitrogen monoxide
- e.  $\text{H}_2\text{Se(aq)}$  hydroselenic acid
- f.  $\text{HNO}_3$  nitric acid
- g.  $\text{HNO}_2$  nitrous acid
- h.  $\text{NaHSO}_4$  sodium hydrogen sulfate

### Half-life

Tritium ( $\text{H-3}$ ) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?



Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?



### Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.



# Chemistry Unit 3 Energy Study Answers

**Léo Pomar**

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**Chemistry Unit 3 Energy Study Answers:**

## **Chemistry Unit 3 Energy Study Answers** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Unit 3 Energy Study Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Table of Contents Chemistry Unit 3 Energy Study Answers**

1. Understanding the eBook Chemistry Unit 3 Energy Study Answers
  - The Rise of Digital Reading Chemistry Unit 3 Energy Study Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit 3 Energy Study Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Unit 3 Energy Study Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit 3 Energy Study Answers
  - Personalized Recommendations
  - Chemistry Unit 3 Energy Study Answers User Reviews and Ratings
  - Chemistry Unit 3 Energy Study Answers and Bestseller Lists

5. Accessing Chemistry Unit 3 Energy Study Answers Free and Paid eBooks
  - Chemistry Unit 3 Energy Study Answers Public Domain eBooks
  - Chemistry Unit 3 Energy Study Answers eBook Subscription Services
  - Chemistry Unit 3 Energy Study Answers Budget-Friendly Options
6. Navigating Chemistry Unit 3 Energy Study Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Unit 3 Energy Study Answers Compatibility with Devices
  - Chemistry Unit 3 Energy Study Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Unit 3 Energy Study Answers
  - Highlighting and Note-Taking Chemistry Unit 3 Energy Study Answers
  - Interactive Elements Chemistry Unit 3 Energy Study Answers
8. Staying Engaged with Chemistry Unit 3 Energy Study Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Unit 3 Energy Study Answers
9. Balancing eBooks and Physical Books Chemistry Unit 3 Energy Study Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Unit 3 Energy Study Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Unit 3 Energy Study Answers
  - Setting Reading Goals Chemistry Unit 3 Energy Study Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Unit 3 Energy Study Answers
  - Fact-Checking eBook Content of Chemistry Unit 3 Energy Study Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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