

**Mastery Properties and Changes Test****Multiple Choice**

Choose the choice that best completes the statement or answers the question.

Choose the item that is an example of the term.

- \_\_\_\_\_ 1. compound  
 a. salt  
 b. air  
 c. metal  
 d. mixture
- \_\_\_\_\_ 2. homogeneous mixture  
 a. oil and vinegar  
 b. air  
 c. metal  
 d. mixture
- \_\_\_\_\_ 3. element  
 a. salt  
 b. brass  
 c. aluminum  
 d. mixture

**Matching**

Match the terms below with their correct definitions.

- |                                |                     |
|--------------------------------|---------------------|
| a. law of multiple proportions | g. crystallization  |
| b. percent by mass             | h. chromatography   |
| c. periodic table              | i. states of matter |
| d. law of conservation of mass | j. solid            |
| e. law of definite proportions | k. liquid           |
| f. solution                    |                     |
- \_\_\_\_\_ 4. A homogeneous mixture
- \_\_\_\_\_ 5. States that mass is neither created nor destroyed in a chemical reaction
- \_\_\_\_\_ 6. States that regardless of the amount, a compound is always composed of the same elements in the same proportion by mass
- \_\_\_\_\_ 7. States that different masses of one element combine with the same relative mass of another element to form different compounds
- \_\_\_\_\_ 8. The grid organizing elements into periods and groups
- \_\_\_\_\_ 9. The ratio of the mass of each element to the total mass of the compound
- \_\_\_\_\_ 10. Three physical forms that describe all matter that exists on Earth
- \_\_\_\_\_ 11. A separation technique that results in the formation of pure solid particles from a solution
- \_\_\_\_\_ 12. A technique that separates the components of a mixture
- \_\_\_\_\_ 13. A form of matter with a constant volume that takes the shape of its container

# Chemistry Standardized Test Practice Answer Key Matter

**Clemens Wendtner**



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