

NAME: _____

Unit 1 Practice Exam 1

1. Atom Structure

Which of the following statements about atomic structure is **NOT** true?

- a) Most of the volume of an atom is empty space.
- b) Electrons are the fundamental particle with a negative charge.
- c) The number of protons and neutrons in an atom must be equal. **correct**
- d) Protons are about 2000 times larger than electrons.

2. Question Functional Relationships

Which of the following statements about functional relationship is **NOT** true?

- a) Plotting (1,2), (2,4), (3,6) is an inverse function. **correct**
- b) Plotting (1,1), (2,4), (3,9) is a parabolic function.
- c) Plotting (1,2), (2,4), (3,8), (4,16) is an exponential function.
- d) Plotting (1,1), (2,0.25), (3,0.11) is an inverse square function.

3. Scientific Notation Math

What is $[(8 \times 10^{-5})(3 \times 10^9)]/(4 \times 10^{-5})$?

- a) 6×10^9 **correct**
- b) 6×10^{-1}
- c) 9.6×10^{10}
- d) 6.6×10^9

4. Approximations

The atomic weights of the halogens are listed below. If you rounded to the nearest integer to perform a calculation of molar mass, which would have the largest error?

- a) F: 19.0 g/mol
- b) Cl: 35.45 g/mol **correct**
- c) Br: 79.9 g/mol
- d) I: 126.9 g/mol

Chemistry Midterm Lab Answers

Elisabeth Fassas



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