

## CHEMICAL BONDING POGIL

### Activity 2 – What's in a name?

#### Objective:

- Identify some simple rules about nomenclature (naming).

#### The Model:

Examine the table below, and answer the following questions.

Table 2

Cation	Anion	Chemical Formula	Compound Name
$\text{Na}^+$	$\text{Cl}^-$	$\text{NaCl}$	sodium chloride
$\text{Ca}^{2+}$	$\text{O}^{2-}$	$\text{CaO}$	calcium oxide
$\text{Zn}^{2+}$	$\text{Cl}^-$	$\text{ZnCl}_2$	zinc chloride
$\text{Li}^+$	$\text{S}^{2-}$	$\text{Li}_2\text{S}$	lithium sulfide
$\text{K}^+$	$\text{N}^{3-}$	$\text{K}_3\text{N}$	potassium nitride

#### Reviewing the Model

- Are ALL cations positive ions or negative ions? *positive*
- Are ALL anions positive ions or negative ions? *negative*
- What is the name of the compound formed by the combination of  $\text{Li}^+$  and  $\text{S}^{2-}$  ions? *Lithium sulfide*

#### Exploring the Model

- When the name of an ionic compound is given, which ion is stated first? *Cation*
- Compare the first part of the compound name to the name of the element from the periodic table. How does the name of the cation correspond to the name of the element?
- Compare the second part of the compound name to the name of the element from the periodic table. How does the name of the anion correspond to the name of the element?
- From what part of the periodic table do the cations in the Model come (metals or nonmetals)?  
*Alkali and alkaline earth*
- From what part of the periodic table do the anions in the Model come?  
*4*

# Chemical Bonding Pogil Answer Key

**NARAYAN CHANGDER**



## **Chemical Bonding Pogil Answer Key:**

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*Chemistry* Richard S. Moog, John J. Farrell, 2017-06-26 In the newly updated 7th Edition Chemistry A Guided Inquiry continues to follow the underlying principles developed by years of extensive research on how students learn and draws on testing by those using the POGIL methodology This text follows the principles of inquiry based learning and correspondingly emphasizes underlying chemistry concepts and the reasoning behind them This text provides an approach that follows modern cognitive learning principles by having students learn how to create knowledge based on experimental data and how to test that knowledge

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**Chemical Bonding and Structure**, 1968

Understanding chemistry Gordon M. Barrow, 1967 METALLIC BOND NARAYAN CHANGDER, 2024-04-05 Note

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