

# Atomic Structure & Periodic Table Worksheet

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1. The table below shows the structure of different atoms and ions. Using the periodic table, complete the table with missing values.

Element	Atomic Number	Proton Number	Neutron Number	Electron Number	Mass Number
Mg	12		12		
Mg <sup>2+</sup>	12				24
F			10		
F <sup>-1</sup>			10		19

2. The table below gives the composition of three different particles X, Y, and Z.

Particle	Proton Number	Neutron Number	Electron Number
X	15	16	15
Y	15	16	18
Z	15	17	15

State the evidence in the table for each of the following statements.

- a) Particle X is an atom.

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- b) Particle Y is a negative ion.

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- c) X, Y, and Z are all particles of the same element.

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- d) Particles X and Z are isotopes of the same elements.

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- e) Particles X and Z have zero charges.

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# Chapter Test Atom And Periodic Table

**M Mark**



## **Chapter Test Atom And Periodic Table:**

## Reviewing **Chapter Test Atom And Periodic Table**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Chapter Test Atom And Periodic Table**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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