

**CHAPTER 4 PRACTICE TEST: SOLUTION STOICHIOMETRY**1) Which of the following are weak electrolytes? HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, NH<sub>3</sub>, KCl

A) HCl, KCl

B) HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, NH<sub>3</sub>, KClC) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, KClD) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, NH<sub>3</sub>E) HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, KCl*Answer: D Diff: 2*2) What are the spectator ions in the reaction between KOH (aq) and HNO<sub>3</sub>(aq)?A) K<sup>+</sup> and H<sup>+</sup>C) K<sup>+</sup> and NO<sub>3</sub><sup>-</sup>E) OH<sup>-</sup> onlyB) H<sup>+</sup> and OH<sup>-</sup>D) H<sup>+</sup> and NO<sub>3</sub><sup>-</sup>*Answer: C Diff: 2*3) The balanced net ionic equation for precipitation of CaCO<sub>3</sub> when aqueous solutions of Na<sub>2</sub>CO<sub>3</sub> and CaCl<sub>2</sub> are mixed is \_\_\_\_\_.A) 2Na<sup>+</sup>(aq) + CO<sub>3</sub><sup>2-</sup>(aq) → Na<sub>2</sub>CO<sub>3</sub>(aq)

E)

B) 2Na<sup>+</sup>(aq) + 2Cl<sup>-</sup>(aq) → 2NaCl(aq)Na<sub>2</sub>CO<sub>3</sub>(aq) + CaCl<sub>2</sub>(aq) → 2NaCl(aq) + CaCO<sub>3</sub>(s)C) Na<sup>+</sup>(aq) + Cl<sup>-</sup>(aq) → NaCl(aq)D) Ca<sup>2+</sup>(aq) + CO<sub>3</sub><sup>2-</sup>(aq) → CaCO<sub>3</sub>(s)*Answer: D Diff: 3*4) Which of the following are strong acids? HI, HNO<sub>3</sub>, HF, HBr

A) HF, HBr

C) HI, HF, HBr

E) HI, HNO<sub>3</sub>, HBrB) HI, HNO<sub>3</sub>, HF, HBrD) HNO<sub>3</sub>, HF, HBr*Answer: E Diff: 2*

5) A neutralization reaction between an acid and a metal hydroxide produces \_\_\_\_\_.

A) water and a salt

C) oxygen gas

E) ammonia

B) hydrogen gas

D) sodium hydroxide

*Answer: A Diff: 2*6) Which of these metals is the least easily oxidized? Na, Au, Fe, Ca, Ag

A) Na

B) Au

C) Fe

D) Ca

E) Ag

*Answer: B Diff: 3*

7) Based on the equations below, which metal is the most active?



A) Ni

B) Ag

C) Cu

D) Pb

E) N

*Answer: A Diff: 3*

8) What is the concentration (M) of KCl in a solution made by mixing 25.0 mL of 0.100 M KCl with 50.0 mL of 0.100 M KCl?

A) 0.100

B) 0.0500

C) 0.0333

D) 0.0250

E) 125

*Answer: A Diff: 3*

# Chapter 7 Chemistry Assessment Answers

**Demosthenes**



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