

## Cells and Energy

**Study Guide B****Answer Key****SECTION 1. CHEMICAL ENERGY AND ATP**

- adenosine triphosphate (ATP)
- a molecule that transfers energy from the breakdown of food molecules to cell processes
- ATP is a high-energy molecule that is converted into lower-energy ADP when a phosphate group is removed and energy is released. ADP is converted back into ATP by the addition of a phosphate group.  
**Cycle Diagram:** High-energy adenosine triphosphate (ATP); phosphate removed, energy released; lower-energy adenosine diphosphate (ADP); energy added from breakdown of carbon-based molecules, phosphate added.
- molecules most commonly broken down; glucose yields about 36 ATP; 4 calories per mg (4-Calories per gram)
- same amount of the energy in a person's body; triglyceride yields about 145 ATP; 9 calories per mg (9-Calories per gram)
- least likely to be broken down; same about the same amount of energy as a mitochondrion; 4 calories per mg (4-Calories per g)
- a process by which some organisms use chemical energy instead of light energy to make energy-storing carbon-based molecules
- ATP has three phosphate groups; ADP has two phosphate groups
- Together, they mean "to put together with chemical energy." In chemosynthesis, chemical energy is used to produce carbon-based molecules that store energy.

**SECTION 2. OVERVIEW OF PHOTOSYNTHESIS**

- they produce the source of chemical energy for themselves and for other organisms
- to capture light energy to make sugars that store chemical energy

- a molecule in chloroplasts that absorbs some of the energy in visible light
- membrane-bound organelles where photosynthesis takes place in plants
- stroma and grana
- coin-shaped, membrane-enclosed compartments inside the grana
- $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$   
Carbon dioxide and water (the reactants) enter photosynthesis. Through many chemical reactions, with the help of many enzymes, a six-carbon sugar and oxygen (the products) are formed.
- The light-dependent reactions require light, and they absorb and transfer energy. The light-independent reactions do not directly need light, and they build sugars.  
**Labels for Chloroplast Steps:** (1) Energy from sunlight is absorbed and transferred along the thylakoid membrane. Water molecules are broken down and oxygen is released. (2) Energy stored along the thylakoid membrane is transferred to molecules that carry energy to the light-independent reactions. (3) Carbon dioxide is added to a cycle of chemical reactions to build larger molecules. (4) A six-carbon simple sugar (usually glucose)  $\text{C}_6\text{H}_{12}\text{O}_6$  is formed.
- Energy from light is used to put sugars together.
- Chlorophyll *a* is the light-absorbing molecule that makes leaves look green.
- The light-dependent reactions require sunlight; the light-independent reactions can occur without sunlight.

**SECTION 3. PHOTOSYNTHESIS IN DETAIL**

- to capture and transfer energy
- groups of molecules that capture and transfer energy

# Biology Chapter One Study Guide Key

**Tao Wei**



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**Study Guide: Sg Concepts in Biology** Eldon D.. Enger, Enger, Frederick C.. Ross, 2003 *Cell and Molecular Biology, Problems Book and Study Guide* Gerald Karp, 2001-09-25 Balances coverage of the concepts of cell and molecular biology using examples of experimentation to support those concepts As experimental techniques become more diverse and complex it is increasingly necessary to identify individual studies that have a broad impact on our understanding of cell biology This text describes in detail some of the key experimental findings along with the original data and figures **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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