

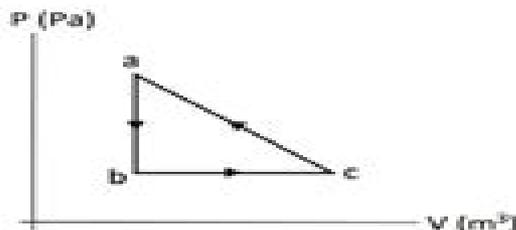
First Law of Thermodynamics

Grade

Name _____ Date _____

Use the following P-V diagram to answer questions about this thermodynamic cycle consisting of three processes labeled ab , bc , and ca .

$P_a = 3.00 \times 10^5 \text{ Pa}$, $P_b = 1.00 \times 10^5 \text{ Pa}$, $V_a = 3 \times 10^{-3} \text{ m}^3$, $V_c = 9 \times 10^{-3} \text{ m}^3$.



a) Identify the type of thermodynamic processes for each part of the cycle (ab , bc , ca) as:

- | | | | |
|--------|----------------|------------------|----------------|
| ab : | i) adiabatic | ii) isobaric | iii) isochoric |
| | iv) isothermal | v) none of these | |
| bc : | i) adiabatic | ii) isobaric | iii) isochoric |
| | iv) isothermal | v) none of these | |
| ca : | i) adiabatic | ii) isobaric | iii) isochoric |
| | iv) isothermal | v) none of these | |

b) For the process ab , determine (i) the work W_{ab} done by the gas and (ii) the change in internal energy of the system for a heat loss of 900 J. [Hints: for (i) recall that W is the area under the curve; for (ii) use the 1st Law.]

c) For the process bc , determine (i) the work W_{bc} done by the gas and (ii) the change in internal energy of the system for a heat gain of 1250 J.

d) For the process ca , determine (i) the work W_{ca} done by the gas and (ii) the change in internal energy of the system for a heat gain of 150 J.

e) Determine the total work W_{tot} done by the gas in the cycle. Compare this to the area of the triangle. Note that both are negative. Also note that the total change in internal energy is zero, as it must be when the system returns to the same state.

Thermodynamic Questions And Answers

J.Selwin Rajadurai



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Thermodynamics and Thermal Engineering J. Selwin Rajadurai, 2003 Thermodynamics And Thermal Engineering A Core Text In SI Units Meets The Complete Requirements Of The Students Of Mechanical Engineering In All Universities Ultimately It Aims At Aiding The Students Genuinely Understand The Basic Principles Of Thermodynamics And Apply Those Concepts To Practical Problems Confidently It Provides A Clear And Detailed Exposition Of Basic Principles Of Thermodynamics Concepts Like Enthalpy Entropy Reversibility Availability Are Presented In Depth And In A Simple Manner Important Applications Of Thermodynamics Like Various Engineering Cycles And Processes Are Explained In Detail Introduction To Latest Topics Are Enclosed At The End Each Topic Is Further Supplemented With Solved Problems Including Problems From Gate Ies Exams Objective Questions Along With Answers Review Questions And Exercise Problems Alongwith Answers For An Indepth Understanding Of The Subject

Fundamentals of Engineering Thermodynamics R. Yadav ,2023-05-10

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Thermal Physics and Statistical Mechanics S. K. Roy, 2001 This Book Emphasises The Development Of Problem Solving Skills In Undergraduate Science And Engineering Students The Book Provides More Than 350 Solved Examples With Complete Step By Step Solutions As Well As Around 100 Practice Problems With Answers Also Explains The Basic Theory Principles Equations And Formulae For A Quick Understanding And Review Can Serve Both As A Useful Text And Companion Book To Those Preparing For Various Examinations In Physics

Thermodynamics with Chemical Engineering Applications Elias I. Franses, 2014-08-25 Master the principles of thermodynamics with this comprehensive undergraduate textbook carefully developed to provide students of chemical engineering and chemistry with a deep and intuitive understanding of the practical applications of these fundamental ideas and principles Logical and lucid explanations introduce core thermodynamic concepts in the context of their measurement and experimental origin giving students a thorough understanding of how theoretical concepts apply to practical situations A broad range of real world applications relate key topics to contemporary issues such as energy efficiency environmental engineering and climate change and further reinforce students understanding of the core material This is a carefully organized highly pedagogical treatment

including over 500 open ended study questions for discussion over 150 varied homework problems clear and objective standards for measuring student progress and a password protected solution manual for instructors

Classical and Quantum Thermal Physics R. Prasad, 2016-11-02 Discusses the interactions of heat energy and matter *The Thermodynamics of Electrical Phenomena in Metals, and A Condensed Collection of Thermodynamic Formulas* Percy Williams Bridgman, 1961 *Schaum's Outline of Thermodynamics for Engineers, Fourth Edition* Merle C. Potter, Craig W. Somerton, 2019-10-22

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Programmed Thermodynamics Charles E. Wales,1970 For the engineering student

Thermodynamics Josiah Willard Gibbs,1906

Foundations of the Thermodynamic Theory of Generalized Fields Mahmoud A. Melehy,1973

The Collected Works of J. Willard Gibbs ...: Thermodynamics Josiah Willard Gibbs,1928

Schaum's Outline of Theory and Problems of Thermodynamics Michael M. Abbott,Hendrick C. Van Ness,1976

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