

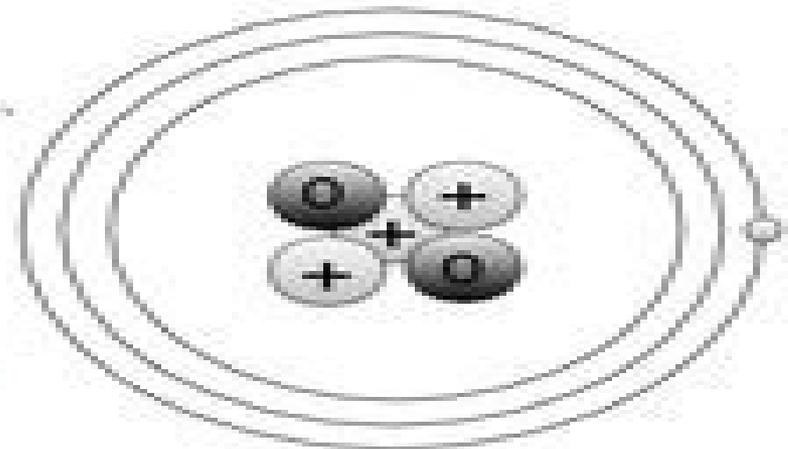
Structure of an Atom Guided Notes

Atoms are the building blocks of **matter**. They are the smallest particle into which an **element** can be **divided** and still retain its identity. An **element** is matter that is made up of only one type of atom.

Atoms have substructure of their own. They have a **nucleus**, which contains **protons** and **neutrons**. Protons have a **positive charge** and a mass of **1** amu (atomic mass unit). Neutrons have **no charge** and are

also measured at 1 amu. The number of protons in the atomic nucleus, which is the **atomic number**, is the defining characteristic of each element.

Electrons, which have a **negative charge** and a mass of almost zero, are located in an area surrounding the nucleus called the **electron cloud**. The way electrons react between charged particles explains the way atoms form **molecules**.



WORD BANK

Matter atomic number electrons molecules no charge positive charge
divided element nucleus protons electron cloud neutrons element

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Table of Contents Guided Notes The Atom

1. Understanding the eBook Guided Notes The Atom
 - The Rise of Digital Reading Guided Notes The Atom
 - Advantages of eBooks Over Traditional Books
2. Identifying Guided Notes The Atom
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Guided Notes The Atom
 - User-Friendly Interface
4. Exploring eBook Recommendations from Guided Notes The Atom
 - Personalized Recommendations
 - Guided Notes The Atom User Reviews and Ratings

- Guided Notes The Atom and Bestseller Lists
- 5. Accessing Guided Notes The Atom Free and Paid eBooks
 - Guided Notes The Atom Public Domain eBooks
 - Guided Notes The Atom eBook Subscription Services
 - Guided Notes The Atom Budget-Friendly Options
- 6. Navigating Guided Notes The Atom eBook Formats
 - ePub, PDF, MOBI, and More
 - Guided Notes The Atom Compatibility with Devices
 - Guided Notes The Atom Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Guided Notes The Atom
 - Highlighting and Note-Taking Guided Notes The Atom
 - Interactive Elements Guided Notes The Atom
- 8. Staying Engaged with Guided Notes The Atom
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Guided Notes The Atom
- 9. Balancing eBooks and Physical Books Guided Notes The Atom
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Guided Notes The Atom
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Guided Notes The Atom
 - Setting Reading Goals Guided Notes The Atom
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Guided Notes The Atom
 - Fact-Checking eBook Content of Guided Notes The Atom
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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