

## Molar Mass Worksheet #1

Calculate the molar masses of the following chemicals:

1)  $\text{Cl}_2$   $2(35.45) = \boxed{70.9 \text{ g/mol}}$

2)  $\text{KOH}$   $39.10 + 16.00 + 1.01 = \boxed{56.11 \text{ g/mol}}$

3)  $\text{BeCl}_2$   $9.01 + 2(35.45) = \boxed{79.91 \text{ g/mol}}$

4)  $\text{FeCl}_3$   $55.85 + 3(35.45) = \boxed{162.2 \text{ g/mol}}$

5)  $\text{BF}_3$   $10.81 + 3(19.00) = \boxed{67.81 \text{ g/mol}}$

6)  $\text{CCl}_2\text{F}_2$   $12.01 + 2(35.45) + 2(19.00) = \boxed{120.91 \text{ g/mol}}$

7)  $\text{Mg(OH)}_2$   $24.31 + 2(16.00) + 2(1.01) = \boxed{58.33 \text{ g/mol}}$

8)  $\text{HF}_6$   $238.03 + 6(19.00) = \boxed{352.03 \text{ g/mol}}$

9)  $\text{SO}_2$   $32.07 + 2(16.00) = \boxed{64.07 \text{ g/mol}}$

10)  $\text{H}_3\text{PO}_4$   $3(1.01) + 30.97 + 4(16.00) = \boxed{98.00 \text{ g/mol}}$

11)  $(\text{NH}_4)_2\text{SO}_4$   $2(14.01) + 8(1.01) + 32.07 + 4(16.00) = \boxed{132.17 \text{ g/mol}}$

12)  $\text{CH}_3\text{COOH}$   $2(12.01) + 4(1.01) + 2(16.00) = \boxed{60.06 \text{ g/mol}}$

13)  $\text{Pb(NO}_3)_2$   $207.2 + 2(14.01) + 6(16.00) = \boxed{331.22 \text{ g/mol}}$

14)  $\text{Ga}_2(\text{SO}_4)_3$   $2(69.72) + 3(32.07) + 9(16.00) = \boxed{379.65 \text{ g/mol}}$

# Calculating Molar Mass Worksheet With Answers

**Clifford Lane**



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## **Table of Contents Calculating Molar Mass Worksheet With Answers**

1. Understanding the eBook Calculating Molar Mass Worksheet With Answers
  - The Rise of Digital Reading Calculating Molar Mass Worksheet With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Calculating Molar Mass Worksheet With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Calculating Molar Mass Worksheet With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Calculating Molar Mass Worksheet With Answers
  - Personalized Recommendations
  - Calculating Molar Mass Worksheet With Answers User Reviews and Ratings
  - Calculating Molar Mass Worksheet With Answers and Bestseller Lists
5. Accessing Calculating Molar Mass Worksheet With Answers Free and Paid eBooks
  - Calculating Molar Mass Worksheet With Answers Public Domain eBooks
  - Calculating Molar Mass Worksheet With Answers eBook Subscription Services
  - Calculating Molar Mass Worksheet With Answers Budget-Friendly Options

6. Navigating Calculating Molar Mass Worksheet With Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Calculating Molar Mass Worksheet With Answers Compatibility with Devices
  - Calculating Molar Mass Worksheet With Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Calculating Molar Mass Worksheet With Answers
  - Highlighting and Note-Taking Calculating Molar Mass Worksheet With Answers
  - Interactive Elements Calculating Molar Mass Worksheet With Answers
8. Staying Engaged with Calculating Molar Mass Worksheet With Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Calculating Molar Mass Worksheet With Answers
9. Balancing eBooks and Physical Books Calculating Molar Mass Worksheet With Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Calculating Molar Mass Worksheet With Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Calculating Molar Mass Worksheet With Answers
  - Setting Reading Goals Calculating Molar Mass Worksheet With Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Calculating Molar Mass Worksheet With Answers
  - Fact-Checking eBook Content of Calculating Molar Mass Worksheet With Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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