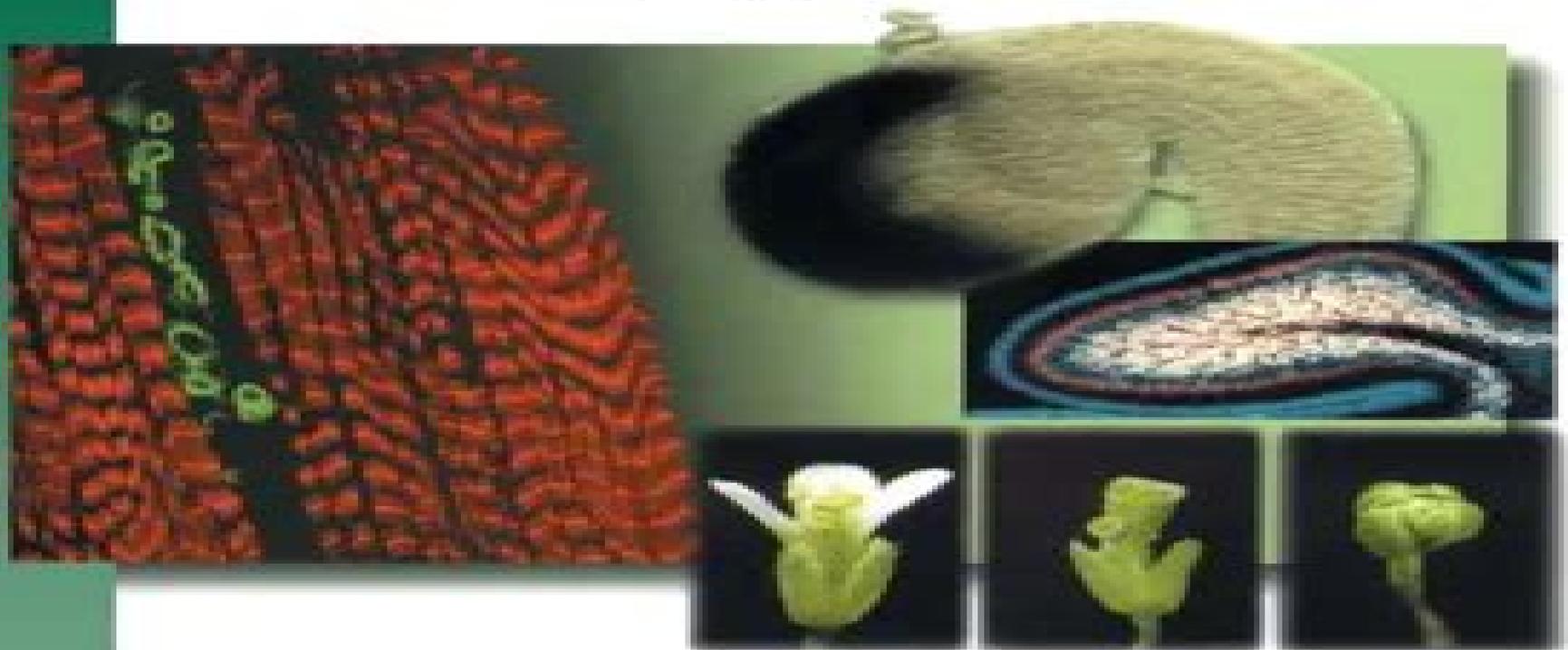


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web when solving a thermodynamic problem follow the following steps read carefully the problem statement draw a picture of the physical situation depicted in the problem statement write in your notebook the givens in the problem statement identify the equations that you will have to use to solve the problem

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web jan 10 2021 using thermodynamic arguments propose an explanation as to why methanol forms conceptual answers in order for the reaction to occur spontaneously  $\Delta g$  for the reaction must be less than zero

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web problem 1 a gas is initially contained inside an insulated container a at initial conditions  $p_1$   $v_1$   $m$  and  $t_1$  these quantities represent pressure volume mass and temperature respectively a valve is then opened which allows the gas to expand freely into an insulated container b which is initially empty

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web thermodynamics problems pdf yuri g melliza processes ideal gas a steady flow compressor handles 113 3 m<sup>3</sup> min of nitrogen  $m$  28 k 1 399 measured at intake where  $p_1$  97 kpa and  $t_1$  27 c discharge is at 311 kpa the changes in

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web in general thermodynamics is concerned with substances in all three phases solid liquid and gas most thermodynamic problems ordinarily involve gases or vapors such as in burning fires though some of thermodynamic problems encountered

may in a few instances involve liquids and solid

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web problem when heating a solution a scientist detects a temperature increase in the solution during a period of time which of the following statements accurately characterizes the solution during this period

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web solution w d 1 5936 0 500mol 8 3145jk1mol1 300k d 1 99 103j q d w d 1 99 103j 3 5 this problem is designed to test the assertion on page 60 that for typical thermodynamic processes in which the elevation of the center of mass changes it is usually a good approximation to set  $w$  equal to  $w_{lab}$

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web solution for a refrigerator cop low  $t$  high  $t$  low solve for the hot side temperature 45 c low 273 high  $t$  low cop 10 45 c 273 250 8k answer is a 2 helium

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web question 1 what is true of isothermal process a  $\Delta t = 0$  b  $\Delta u = 0$  c  $\Delta q = \Delta w$  d  $p, v$  constants solution in an isothermal process temperature remains constant  $\Delta t = 0$  since internal energy depends on the temperature  $\Delta u = 0$  from first law of thermodynamics  $\Delta u = \Delta q + \Delta w$  since  $\Delta u = 0$   $\Delta q = -\Delta w$  also  $p, v, n, r, t$  as  $t$  is constant  $p, v$