

Charge Calculation Exercise

A foundry needs to produce a heat of 2 Tons of 6061 alloy. They have prepared a melt with 80% of total required tonnage using raw materials in foundry. Composition of this melt is given in following table:

Element	Mg	Si	Iron	Cu	Zn	Ti	Mn	Cr	Al
Wt.%	0.5	0.3	0.5	0.1	0.2	0.1	0.1	0.2	rest

1. Search for required elemental composition (wt.%) of 6061.
2. Then Calculate the required amounts of master alloys that need to be added to obtain the required wt.%. Present data in tabular form.
3. You should also comment on addition method of these master alloys after reading some literature.
4. Three bonus marks are for neatness and proper management of the assignment.

Note: Available master alloys of Aluminum in this foundry are:

Mg15%; Cr5%; Mn10%; Si20%; Fe2%; Cu5%

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Table of Contents Foundry Charge Calculation

1. Understanding the eBook Foundry Charge Calculation
 - The Rise of Digital Reading Foundry Charge Calculation
 - Advantages of eBooks Over Traditional Books
2. Identifying Foundry Charge Calculation
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Foundry Charge Calculation
 - User-Friendly Interface
4. Exploring eBook Recommendations from Foundry Charge Calculation
 - Personalized Recommendations
 - Foundry Charge Calculation User Reviews and Ratings
 - Foundry Charge Calculation and Bestseller Lists
5. Accessing Foundry Charge Calculation Free and Paid eBooks

- Foundry Charge Calculation Public Domain eBooks
 - Foundry Charge Calculation eBook Subscription Services
 - Foundry Charge Calculation Budget-Friendly Options
6. Navigating Foundry Charge Calculation eBook Formats
 - ePub, PDF, MOBI, and More
 - Foundry Charge Calculation Compatibility with Devices
 - Foundry Charge Calculation Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Foundry Charge Calculation
 - Highlighting and Note-Taking Foundry Charge Calculation
 - Interactive Elements Foundry Charge Calculation
 8. Staying Engaged with Foundry Charge Calculation
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Foundry Charge Calculation
 9. Balancing eBooks and Physical Books Foundry Charge Calculation
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Foundry Charge Calculation
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Foundry Charge Calculation
 - Setting Reading Goals Foundry Charge Calculation
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Foundry Charge Calculation
 - Fact-Checking eBook Content of Foundry Charge Calculation
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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