

## Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 2

**Q1.** What is the mass of one atom of C – 12 in grams?

- a.)  $1.992 \times 10^{-23}$  g
- b.)  $1.989 \times 10^{-23}$  g
- c.)  $1.992 \times 10^{-23}$  g
- d.)  $1.965 \times 10^{-23}$  g

**Q2.** Which of the following laws states that the total mass of the reactants is equal to the total mass of the products in a chemical reaction?

- a.) Law of multiple proportions
- b.) Law of conservation of mass
- c.) Avogadro's law
- d.) Law of constant proportions

**Q3.** The law of multiple proportions was stated by:

- a.) Dalton
- b.) Proust
- c.) Avogadro
- d.) None of the above

**Q4.** The units  $\text{kg m}^{-1} \text{s}^{-2}$  are of \_\_\_\_\_

- a.) Force
- b.) Pressure
- c.) Acceleration
- d.) Velocity

**Q5.** The empirical formula of glucose is \_\_\_\_\_.

- a.)  $\text{CH}_2\text{O}$
- b.)  $\text{CHO}_2$
- c.)  $\text{C}_2\text{H}_2\text{O}$
- d.)  $\text{CH}_2\text{O}_2$

**Q6.** Express the following in the scientific notation:

- i.) 0.0048

# Multiple Choice Solution Chemistry 11 Questions

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## **Multiple Choice Solution Chemistry 11 Questions** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Multiple Choice Solution Chemistry 11 Questions**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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