

Ideal gas law - problems and solutions

1. Ideal gases in a closed container initially have volume V and temperature T . The final temperature is $5/4T$ and the final pressure is $2P$. What is the final volume of gases.

Known:

Initial volume (V_1) = V

Initial temperature (T_1) = T

Final temperature (T_2) = $5/4 T$

Initial pressure (P_1) = P

Final pressure (P_2) = $2P$

Wanted: Final volume (V_2)

Solution:

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$\frac{P V}{T} = \frac{(2P) V_2}{5/4 T}$$

$$\frac{V}{1} = \frac{2 V_2}{5/4}$$

$$5/4 V = 2 V_2$$

$$V_2 = \frac{5V}{4} \cdot \frac{2}{1}$$

$$V_2 = \frac{5V}{4} \times \frac{1}{2}$$

$$V_2 = \frac{5V}{8}$$

2. Determine the volume of 2.00 moles of gases (ideal gas) at STP. STP = Standard Temperature and Pressure.

Known:

Moles of gas (n) = 2 moles

Standard temperature (T) = $0^\circ\text{C} = 0 + 273 = 273$ Kelvin

Standard pressure (P) = 1 atm = 1.013×10^5 Pa

Universal gas constant (R) = 8.315 Joule/mole.Kelvin

Wanted: Volume of gases (V)

Solution:

Read more:

<https://physics.gurumuda.net/ideal-gas-law-problems-and-solutions.htm>

Gas Law Problems With Solutions

Silberberg

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Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers Questions and Answers on Federal Tax Laws, Problems Affecting Individuals, Partnerships and Corporations, January, 1924 Irving Bank-Columbia Trust Company,1924 Physico-chemical Calculations Joseph Knox,1919 *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **University High School Journal** ,1925 *Student Solutions Manual for Chang's Chemistry* Philip C. Keller,Raymond Chang,Jill L. Keller,1988 **Journal** University High School (Oakland, Calif.),1926 *Physico-chemical Calculations* Joseph Knox,1944 Chemical Calculations Raymond Harman Ashley,1929 **eBook: General, Organic and Biological Chemistry 2e** SMITH,2012-02-16 eBook General Organic and Biological Chemistry 2e Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Journal of the American Chemical Society** American Chemical Society,1921 Proceedings of the Society are included in v 1 59 1879 1937 **Chemical News and Journal of Industrial Science** ,1908 *The Chemical News and Journal of Physical Science* ,1913 *The Journal of Physical Chemistry* Wilder Dwight Bancroft,1912 Includes section New Books *The Journal of Physical Chemistry* ,1912 *The Chemical News and Journal of Industrial Science* William Crookes,James H. Gardiner,Gerald Druce,H. W. Blood-Ryan,1908

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