

Name: \_\_\_\_\_

Lab Instructor: \_\_\_\_\_

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**PREPARATION FOR CHEMISTRY LAB: MEASUREMENT (Part I)**

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Pre-lab questions need to be completed prior to your coming to lab. They may be collected at the beginning of the laboratory period.

1. What are the units of mass, length, and temperature in the International System of Units (Table 1.2 in your textbook)?
2. Perform the following calculation and report the answer to the correct number of significant figures (assume all quantities are measured):  
 $4.11 + 8.9 + 0.1060$
3. Assuming all numbers come from measurements, perform the following calculation and report the answer to the correct number of significant figures  
 $2.415 \times 8.6 \times (2.08 \times 10^5)$
4. Calculate the number of mm that are in 538.3 inches. Use  $2.54 \text{ cm} = 1 \text{ in}$  and show your work.
5. If you have  $3.6 \text{ cm}^3$  of water, how many liters of water do you have?

# Preparation For Chemistry Lab Measurement Part I Number

**RC Schank**



**Preparation For Chemistry Lab Measurement Part I Number:**

## Decoding **Preparation For Chemistry Lab Measurement Part I Number**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Preparation For Chemistry Lab Measurement Part I Number**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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