



10. Let  $x$ ,  $y$  and  $z$  be positive real numbers. Suppose  $x$ ,  $y$  and  $z$  are the lengths of the sides of a triangle opposite to its angles  $X$ ,  $Y$  and  $Z$ , respectively. If

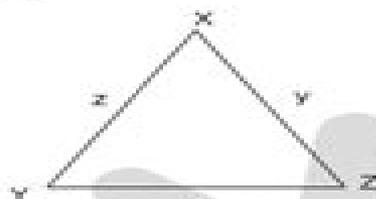
$$\tan \frac{X}{2} + \tan \frac{Z}{2} = \frac{2y}{x+y+z}$$

then which of the following statements is/are TRUE?

- (A)  $2Y = X + Z$
- (B)  $Y = X + Z$
- (C)  $\tan \frac{X}{2} = \frac{x}{y+z}$
- (D)  $x^2 + z^2 - y^2 = xz$

Answer: B, C

Solution:



$$\tan \frac{X}{2} + \tan \frac{Z}{2} = \frac{2y}{x+y+z}$$

$$\Rightarrow \frac{\Delta}{S(S-x)} + \frac{\Delta}{S(S-z)} = \frac{2y}{2S}$$

$$\Rightarrow \frac{\Delta}{S} \left( \frac{2S - (x+z)}{(S-x)(S-z)} \right) = \frac{y}{S}$$

$$\Rightarrow \frac{\Delta y}{S(S-x)(S-z)} = \frac{y}{S}$$

$$\Rightarrow \Delta^2 = (S-x)^2(S-z)^2$$

$$\Rightarrow S(S-y) = (S-x)(S-z)$$

$$\Rightarrow (x+y+z)(x+z-y) = (y+z-x)(x+y-z)$$

$$\Rightarrow (x+z)^2 - y^2 = y^2 - (z-x)^2$$

# Jee Advanced Model Test Papers

**Didier Musso**



## **Jee Advanced Model Test Papers:**

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oct 10 2017 senyawa halogen reaksi halogen dengan logam halogen bereaksi dengan sebagian besar logam menghasilkan senyawa garam na cl<sub>2</sub> nacl 2fe 3cl<sub>2</sub> 2fecl<sub>3</sub> sn 2cl<sub>2</sub> sncl<sub>4</sub> mg cl<sub>2</sub> mgcl<sub>2</sub> 2al 3cl<sub>2</sub> 2alcl<sub>3</sub> 2b 3cl<sub>2</sub> 2bcl<sub>3</sub> 2si 2cl<sub>2</sub> sicl<sub>4</sub> 2k cl<sub>2</sub> 2kcl

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gas mulia adalah unsur unsur yang terdapat dalam golongan viiia yang memiliki kestabilan yang sangat tinggi dan sebagian ditemukan di alam dalam bentuk monoatomik karena sifat stabilnya unsur unsur yang terdapat dalam gas mulia yaitu helium he neon ne argon ar kripton kr xenon xe radon rn

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oct 16 2023 kerapatan densitas unsur unsur gas mulia juga akan semakin bertambah dari atas ke bawah sifat fisik golongan viiia gas mulia sulit bereaksi bersifat inert karena konfigurasi elektronnya stabil sehingga jarang ditemukan dalam bentuk senyawa gas mulia sedikit larut dalam air kecuali helium dan neon karena ukuran atomnya terlalu kecil

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gas mulia ditemukan di dekat golongan halogen dalam tabel periodik karena unsur gas mulia memiliki konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas gas ini dikenal dengan gas inert

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gas mulia gas mulia adalah unsur unsur yang terdapat dalam golongan viiia yang memiliki kestabilan yang sangat tinggi dan sebagian ditemukan di alam dalam bentuk monoatomik unsur unsur yang terdapat dalam gas mulia yaitu helium he neon ne argon ar kripton kr xenon xe radon rn

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sifat sifat gas mulia wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik molekul yg terdiri dari 1 atom ini erat kaitannya dengan struktur elektron oktet dan duplet dari gas mulia

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dec 3 2015 gas mulia dan halogen merupakan unsur kimia yang sifatnya sangat bertolak belakang gas mulia dinamakan mulia karena seperti logam mulia sukar sekali bereaksi sementara halogen halos garam artinya adalah pembentuk garam halogen memiliki 6 bilangan oksidasi sehingga mudah sekali berreaksi dengan unsur lainnya gas mulia noble

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pembuatan gas mulia gas helium mempunyai titik didih yang sangat rendah yaitu 268 8 0c sehingga pemisahan gas helium

dari gas alam dilakukan dengan cara pengembunan liquefaction sampai gas alam akan mencair sekitar 156 Oc dan gas helium terpisah dari gas alam pembuatan ne ar kr dan xe dilakukan dengan proses destilasi bertingkat

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golongan gas mulia dan halogen beberapa golongan unsur memiliki nama khusus seperti golongan viiia gas mulia dan viia halogen golongan tersebut memiliki nama khusus seperti sifat priodik sifat fisik dan sifat kimia kekhas sifat ke 2 golongan tersebut dapat membantu kita unsur unsur yang lain dalam 1 golongan semakin ke atas dan dalam

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oct 7 2021 disebut gas mulia karena semua unsur pada golongan ini berwujud gas dan memiliki konfigurasi elektron yang sangat stabil sehingga akan sangat sulit untuk bereaksi dengan unsur lainnya unsur unsur gas mulia antara lain adalah helium he neon ne argon ar kripton kr xenon xe dan radon rn

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nov 26 2015 1 wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik ini erat kaitannya dengan struktur elektronoktet dan duplet dari gas mulia sedangkan wujud gas pada suhu kamar disebabkan titik

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konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas gas ini dikenal dengan gas inert

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