

Name : \_\_\_\_\_ Date : \_\_\_\_\_

## Naming Ionic Compounds Practice Worksheet

Name the following ionic compounds.

1.  $\text{NH}_4\text{Cl}$  \_\_\_\_\_
2.  $\text{Fe}(\text{NO}_3)_2$  \_\_\_\_\_
3.  $\text{TiBr}_3$  \_\_\_\_\_
4.  $\text{Cu}_3\text{P}$  \_\_\_\_\_
5.  $\text{SnSe}_2$  \_\_\_\_\_
6.  $\text{GaAs}$  \_\_\_\_\_
7.  $\text{Pb}(\text{SO}_4)_2$  \_\_\_\_\_
8.  $\text{Be}(\text{HCO}_3)_2$  \_\_\_\_\_
9.  $\text{Mn}_2(\text{SO}_3)_3$  \_\_\_\_\_
10.  $\text{Al}(\text{CN})_3$  \_\_\_\_\_
11.  $\text{LiCl}$  \_\_\_\_\_
12.  $\text{Na}_2\text{O}$  \_\_\_\_\_
13.  $\text{Cr}_3(\text{PO}_4)_2$  \_\_\_\_\_
14.  $\text{LiI}$  \_\_\_\_\_
15.  $\text{V}(\text{CO}_3)_2$  \_\_\_\_\_
16.  $\text{Sn}(\text{NO}_2)_2$  \_\_\_\_\_
17.  $\text{CoCO}_3$  \_\_\_\_\_
18.  $\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2$  \_\_\_\_\_
19.  $\text{Pb}_3\text{N}_2$  \_\_\_\_\_
20.  $\text{V}_2\text{S}_5$  \_\_\_\_\_

# Naming Ionic Compounds Worksheet Answers

**Michael Brown**



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