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Alkanes

- Alkanes are a type of hydrocarbon with single bonds and saturated.
- They all have a general molecular formula of $C_n H_{2n+2}$
- In the structural formula keep in mind they have single bonds
- Empirical formula is not the simplest ratio.
- A list of the different Alkanes

Table 18.1 The First Ten Straight-Chain Alkanes

NAME	MOLECULAR FORMULA	STRUCTURAL FORMULA
methane	CH_4	CH_4
ethane	C_2H_6	CH_3-CH_3
propane	C_3H_8	$CH_3-CH_2-CH_3$
butane	C_4H_{10}	$CH_3-CH_2-CH_2-CH_3$
pentane	C_5H_{12}	$CH_3-CH_2-CH_2-CH_2-CH_3$
hexane	C_6H_{14}	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_3$
heptane	C_7H_{16}	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$
octane	C_8H_{18}	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$
nonane	C_9H_{20}	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$
decane	$C_{10}H_{22}$	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$

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