

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ a. CH_3COOH
- _____ b. Lewis Acids
- _____ c. Carbonyl Compounds
- _____ d. Alcohols
- _____ e. Metal Ions
- _____ f. Simple Molecules
- _____ g. Molecules with Lone Pairs
- _____ h. Lewis Acids
- _____ i. H^+ Ions

- _____ j. Molecules that are Lewis Acids
- _____ k. Molecules that are Lewis Bases
- _____ l. Molecules that are Lewis Acids and Bases
- _____ m. Molecules that are Lewis Acids and Bases
- _____ n. Molecules that are Lewis Acids and Bases
- _____ o. Molecules that are Lewis Acids and Bases
- _____ p. Molecules that are Lewis Acids and Bases
- _____ q. Molecules that are Lewis Acids and Bases
- _____ r. Molecules that are Lewis Acids and Bases
- _____ s. Molecules that are Lewis Acids and Bases
- _____ t. Molecules that are Lewis Acids and Bases
- _____ u. Molecules that are Lewis Acids and Bases
- _____ v. Molecules that are Lewis Acids and Bases
- _____ w. Molecules that are Lewis Acids and Bases
- _____ x. Molecules that are Lewis Acids and Bases
- _____ y. Molecules that are Lewis Acids and Bases
- _____ z. Molecules that are Lewis Acids and Bases

Multiple Choice

2. The Lewis structure of CO_2 is:

- _____ a. $\text{O}=\text{C}=\text{O}$
- _____ b. $\text{O}=\text{C}=\text{O}$
- _____ c. $\text{O}=\text{C}=\text{O}$
- _____ d. $\text{O}=\text{C}=\text{O}$

3. The Lewis structure of CO_2 is: $\text{O}=\text{C}=\text{O}$. The Lewis structure of CO_2 is: $\text{O}=\text{C}=\text{O}$.

- _____ a. $\text{O}=\text{C}=\text{O}$
- _____ b. $\text{O}=\text{C}=\text{O}$
- _____ c. $\text{O}=\text{C}=\text{O}$
- _____ d. $\text{O}=\text{C}=\text{O}$

4. The Lewis structure of CO_2 is: $\text{O}=\text{C}=\text{O}$. The Lewis structure of CO_2 is: $\text{O}=\text{C}=\text{O}$.

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Lewis R. Aiken



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