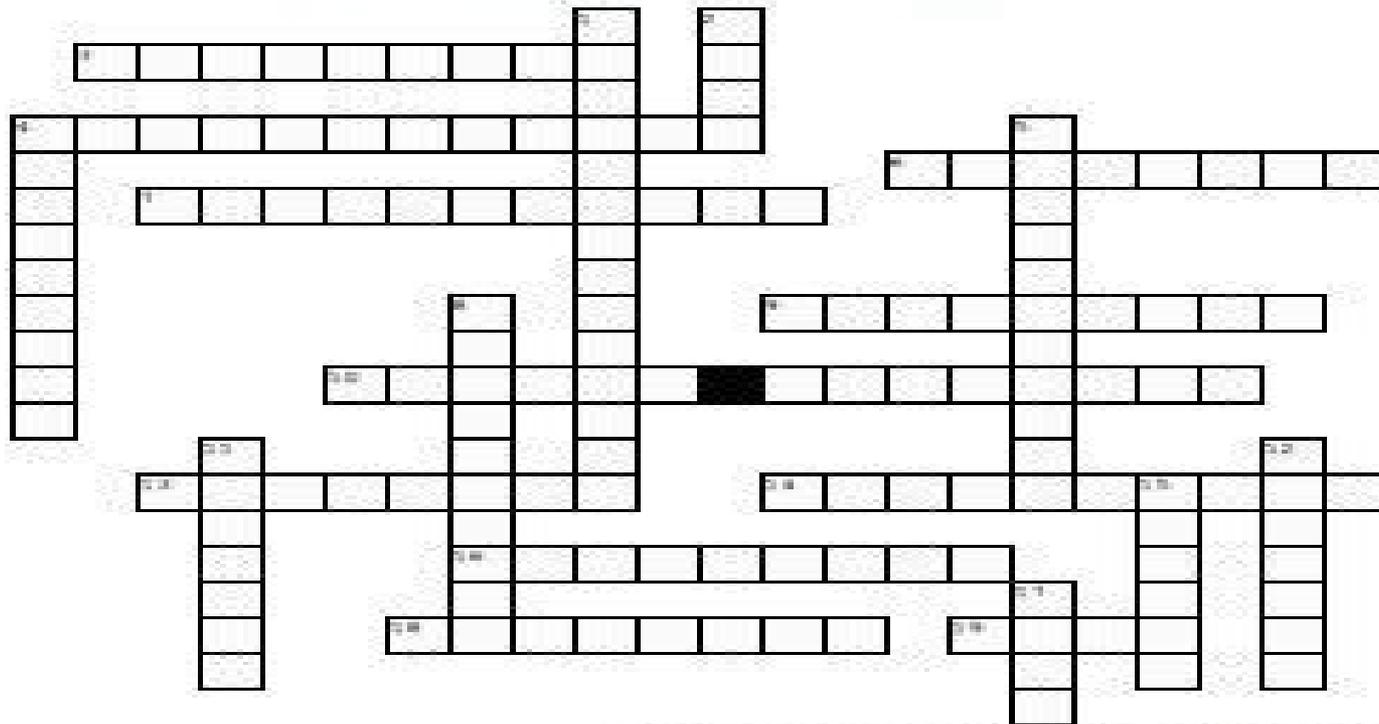


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Across

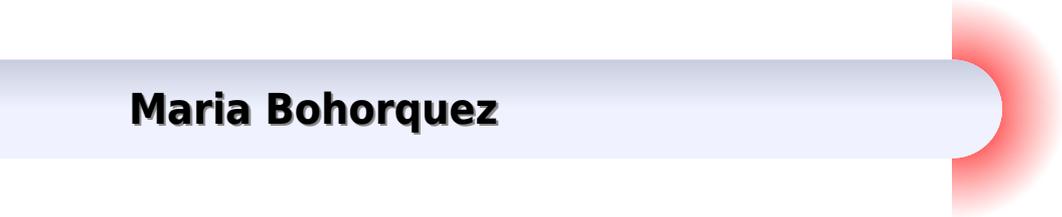
1. The strongest bases release this ion in water.
4. Describes a solution where a lot of solute is dissolved in the water.
5. Solute lowers the _____ point of a solvent.
7. Solution remains this way if you can continue to dissolve more solute.
8. Solution becomes this when so much solute is added that no more dissolves.
10. A solution where only a little solute is dissolved in the water.
13. Is a well-mixed mixture.
14. Is a mixture in which particles can be seen and easily separated by settling or filtration.
15. A compound that changes color when in contact with an acid or a base.
18. Most acids release this ion in water.
19. Any ionic compound made from a neutralization of an acid with a base.

Down

1. This type of acid-base reaction produces water and a salt.
2. Is a substance that tastes sour, reacts with metals, and turns blue litmus paper red.
4. Property of acids that describes their ability to "eat away" at other materials.
6. One factor that affects solubility (hint: can be measured in degrees Celsius).
8. Is a measure of how well a solute can dissolve in a solvent at a given temperature.
11. Solute raises the _____ point of a solvent.
12. The type of _____ also affects solubility, whether it's polar or non-polar.
15. The part of a solution that is present in a solution in a smaller amount and dissolved by the solvent.
17. Is a substance that tastes bitter, feels slippery, and turns red litmus paper blue.

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Maria Bohorquez



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