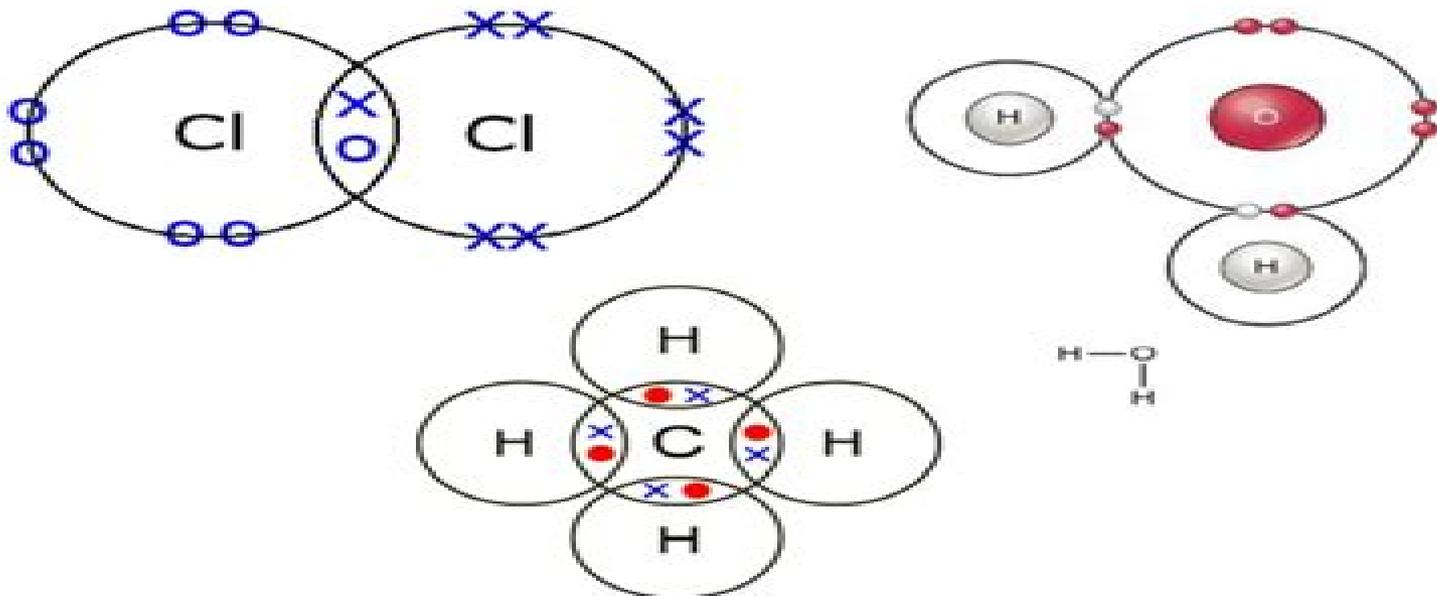


## COVALENT BONDING

- Chemical bond formed by sharing of electrons (from the valence shell) between two atoms, so that each atom in the resulting molecule gets the stable electronic configuration of the nearest noble gas.
- A covalent bond is formed only when both reacting atoms need electrons to achieve stable electronic configuration. ie, Non-metal atoms form covalent bonds
- When two non-metal atoms come close enough, their shells overlap and they share electrons. The strong force of attraction developed between the atoms when they share electrons is called covalent bond.
- Three types of covalent bond – Single , Double , Triple – based on the number of electrons shared by each non-metal atom pair.



# Chapter 8 Covalent Bonding Notes Davis School

**L Cohen**



## **Chapter 8 Covalent Bonding Notes Davis School:**

## Unveiling the Magic of Words: A Overview of "**Chapter 8 Covalent Bonding Notes Davis School**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Chapter 8 Covalent Bonding Notes Davis School**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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