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Acid correct answers Compound that releases hydrogen ions (H^+) in solution

Activation Energy correct answers Amount of energy greater than the energy contained in the reactants, which must be overcome for a reaction to proceed

Adenosine Triphosphate (ATP) correct answers Nucleotide containing ribose and an adenine base that is essential in energy transfer

Amino Acid correct answers Building block of proteins; characterized by an amino and carboxyl functional groups and a variable side-chain

Anion correct answers Atom with a negative charge

Atom correct answers Smallest unit of an element that retains the unique properties of that element

Atomic Number correct answers Number of protons in the nucleus of an atom

Base correct answers Compound that accepts hydrogen ions (H^+) in solution

Bond correct answers Electrical force linking atoms

Buffer correct answers Solution containing a weak acid or a weak base that opposes wide fluctuations in the pH of body fluids

Carbohydrate correct answers Class of organic compounds built from sugars, molecules containing carbon, hydrogen, and oxygen in a 1-2-1 ratio

Catalyst correct answers Substance that increases the rate of a chemical reaction without itself being changed in the process

Cation correct answers Atom with a positive charge

Chemical Energy correct answers Form of energy that is absorbed as chemical bonds form, stored as they are maintained, and released as they are broken

Colloid correct answers Liquid mixture in which the solute particles consist of clumps of molecules large enough to scatter light

Compound correct answers Substance composed of two or more different elements joined by chemical bonds

Anatomy And Physiology Chapter 2

Xiaolong Qi



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