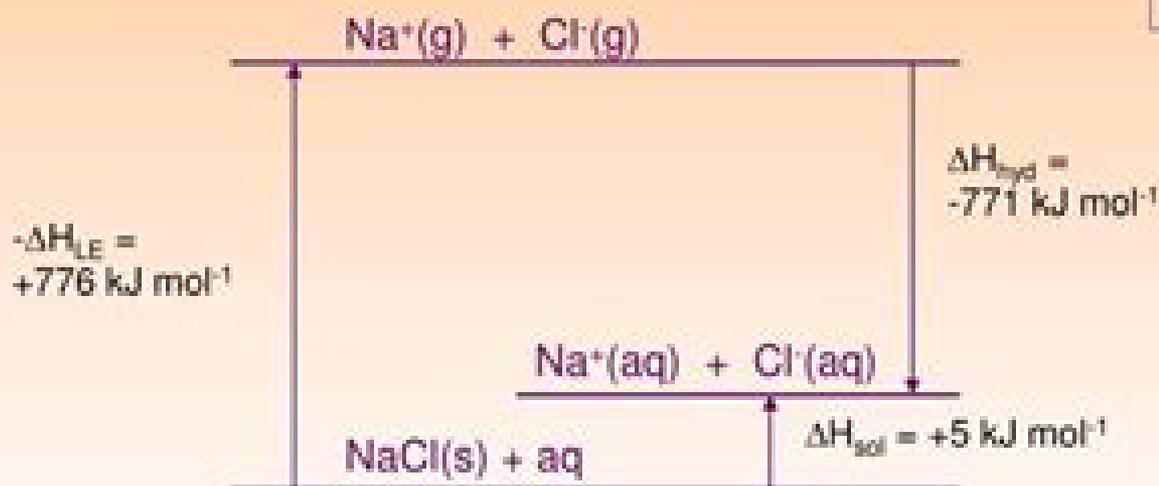


# Enthalpy of Solution

- Enthalpy Diagram

Click to complete the diagram.



$$\Delta H_{sol} = -\Delta H_{LE} + (\Delta H_{hyd(cation)} + \Delta H_{hyd(anion)})$$

$$\Delta H_{sol} = -776 + -771 = +5 \text{ kJ mol}^{-1}$$

# Enthalpy Of Solution Cacl2

**S Baum**



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