

## O level Chemistry Notes



**Atoms:** Building block of matter. It may or may not exist free in nature

Examples: H, He, Na, C



**Nucleus:** It is central part of an atom.

**Shells:** Circular paths around the nucleus where electrons reside. They are also called orbits or energy levels.

No. of es in a shell =  $2n^2$

Where n = no of shell.

1<sup>st</sup> shell = 2es, 2<sup>nd</sup> shell = 8es, 3<sup>rd</sup> shell = 18es.

At least 8 electron should be present in last for the stability of an atom. However if an atom has one shell like Hydrogen and Helium then 2 electrons are required for complete shell and stability.

**Atomic number:**

Number of protons present in the nucleus of an atom of an element.

Examples:  $_{11}\text{Na}$ ,  $_6\text{C}$ ,  $_8\text{O}$

**Atomic mass or Mass number or nucleon number:** It is the sum of protons and neutrons present in the nucleus of an atom of an element.

Examples:  $^{23}\text{Na}$ ,  $^{12}\text{C}$ ,  $^{16}\text{O}$

# O Level Chemistry Notes Studyguide Pk

**R Pring**



## **O Level Chemistry Notes Studyguide Pk:**

## Unveiling the Magic of Words: A Review of "**O Level Chemistry Notes Studyguide Pk**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**O Level Chemistry Notes Studyguide Pk**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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