

Chapter Test B

Chapter: Chemical Equations and Reactions

Match a name space (possibly writing the name of the react or product molecule) completed with reactants/products arrows with equation.

1. The product(s) of a slightly soluble salt decomposed by a double displacement reaction is/are the formation of a
 a. gas

a. precipitate
 b. colorless solution
 c. halogen

2. An ionic compound aqueous is decomposed by adding the
a. precipitate

b. solution
 c. structure of the particles
 d. number of particles

3. An ionic compound aqueous that is soluble in water provides the following ions
 a. water

b. dissolved in water
 c. insoluble
 d. insoluble

4. The resulting ion decomposition reaction on water decomposition is/are
 a. chemical decomposition and reduction for the resulting half-reactions

b. decomposition structure
 c. water decomposition reaction
d. chemical decomposition and reduction for the resulting half-reactions

5. The conditions for chemical reactions
a. indicate the number of moles of each substance that react

b. show the number of grams of each substance that react
 c. indicate the number of moles of the substances
 d. show the relative proportions of each atom

6. If x and y are linear functions of t with x and y variables, then
 a. x will replace y in all the relations

b. y will replace x in all the relations
 c. y will replace x only in the relations
 d. x will replace y only in the relations

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