

# Naming Polyatomic Ionic Compounds Worksheet

Name the following ions.

1)  $(\text{NH}_4)_2\text{CrO}_4$  \_\_\_\_\_

2)  $\text{AgNO}_3$  \_\_\_\_\_

3)  $\text{Al}(\text{ClO}_3)_3$  \_\_\_\_\_

4)  $\text{Ba}(\text{ClO}_4)_2$  \_\_\_\_\_

5)  $\text{CaSO}_4$  \_\_\_\_\_

6)  $\text{Cd}(\text{OH})_2$  \_\_\_\_\_

7)  $\text{Cu}(\text{ClO}_4)_2$  \_\_\_\_\_

8)  $\text{Fe}(\text{HSO}_3)_2$  \_\_\_\_\_

9)  $\text{Fe}(\text{NO}_2)_2$  \_\_\_\_\_

10)  $\text{Hg}_2(\text{NO}_3)_2$  \_\_\_\_\_

11)  $\text{HgCrO}_4$  \_\_\_\_\_

12)  $\text{K}_2\text{Cr}_2\text{O}_7$  \_\_\_\_\_

13)  $\text{Mg}_3(\text{PO}_4)_2$  \_\_\_\_\_

14)  $\text{Pb}(\text{CO}_3)_2$  \_\_\_\_\_

15)  $\text{Pb}(\text{SO}_4)_2$  \_\_\_\_\_

16)  $\text{Pb}_3(\text{PO}_4)_2$  \_\_\_\_\_

17)  $\text{Sn}(\text{CN})_2$  \_\_\_\_\_

18)  $\text{Sn}(\text{CO}_3)_2$  \_\_\_\_\_

19)  $\text{Sr}(\text{NO}_3)_2$  \_\_\_\_\_

20)  $\text{Zn}(\text{HCO}_3)_2$  \_\_\_\_\_

# Chemistry Ionic Compounds Polyatomic Ions Worksheet Answers

**M Mosston**



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