

CHEMICAL BONDING POGIL

Activity 3 – What's in a name?

Objective:

- Identify some simple rules about nomenclature (naming).

The Model:

Examine the table below, and answer the following questions.

Table 2

Cation	Anion	Chemical Formula	Compound Name
Na^+	Cl^-	NaCl	sodium chloride
Ca^{2+}	O^{2-}	CaO	calcium oxide
Zn^{2+}	Cl^-	ZnCl_2	zinc chloride
Li^+	S^{2-}	Li_2S	lithium sulfide
K^+	N^{3-}	K_3N	potassium nitride

Reviewing the Model

- Are ALL cations positive ions or negative ions? *positive*
- Are ALL anions positive ions or negative ions? *negative*
- What is the name of the compound formed by the combination of Li^+ and S^{2-} ions? *Lithium sulfide*

Exploring the Model

- When the name of an ionic compound is given, which ion is stated first?
Cation
- Compare the first part of the compound name to the name of the element from the periodic table. How does the name of the cation correspond to the name of the element?
- Compare the second part of the compound name to the name of the element from the periodic table. How does the name of the anion correspond to the name of the element?
- From what part of the periodic table do the cations in the Model come (metals or nonmetals)?
Alkali and alkaline earth
- From what part of the periodic table do the anions in the Model come?
4

Chemical Bonding Pogil Answers Key

Henry A. Bent



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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more *CHEMICAL BONDING* NARAYAN CHANGDER, 2024-03-31 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more *Chemical Bonding* Audrey L. Companion, 1979 **Atomic Structure and Chemical Bonding, a Non-mathematical Introduction** Fritz Seel, 1963 **Chemical Bonding [electronic Resource]**, 1991 **Chemical Bonding** M.S. Sethi & M. Satake, 2010 Contents Chemical Bonding I Basic Concepts Chemical Bonding II Additional Aspects Intermolecular Force and Crystal Structures [How Chemical Bonds Form and Chemical Reactions Proceed](#) V. I[U] Gankin, Yurii V. Gankin, 1998 **Chemical Bonding** Kok Koon Leong, 2021-05-03 This book introduces the principles behind chemical bonding to teenagers between the ages of fifteen to seventeen Topics covered include ionic bonding covalent bonding and metallic bonding [Atoms & Chemical Bonding Science Learning Guide](#) NewPath Learning, 2014-03-01 The Atoms Atomic Configuration Chemical Bonding Ionic Bonding Ionic Compounds Covalent Bonding Covalent Compounds Naming Compounds and Metallic Bonding Aligned to Next Generation Science Standards NGSS and other state standards **Teaching Chemical Bonding** Margaret Irene Lindsay, 1995 This document presents an instructional strategy for teaching chemical bonding using parables and music Games student interactions and worksheets

are included in the lesson plans Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding JRH **The Answer Key** Rachel Turoscy,2018-08-09 The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to problem solving methodology and unit conversions In subseq **Chemical Bonding - Types, Nature, Characteristics (Chemistry Quick Facts)** E Staff, Learn and review on the go Use Quick Review Chemistry Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Review all the important facts you need to know about various chemical reactions nature characteristics and more Perfect study notes for all high school health sciences premed medical and nursing students **Chemical Bonding** Mark J. Winter,2016 *Chemical Bonding* M. S. Sethi,M. Satake,1992 **Lectures on Chemical Bonding and Quantum Chemistry** Sambhu N. Datta,2019 The concept of a chemical bond evolved from a variety of experimental observations It became useful to understand at times even predict the molecular structure reactivity and mechanism of chemical reactions Every aspect of the concept of bonding received a quantitative interpretation from the advent of quantum mechanics and its application to chemistry In Lectures on Chemical Bonding and Quantum Chemistry the reader will find a comprehensive discourse on the basic interpretation of the chemical bond as well as current understanding in terms of a dancing molecule that not only travels rotates and pulsates around an equilibrium molecular structure but also interacts and collides with other molecules thereby transferring linear and angular momentum characteristics and adjusting total energies One will also find a thorough survey of quantum mechanical methodologies for calculation of molecular characteristics in specific states and their changes under spectroscopic transitions tunneling electron and proton transfer phenomena and so on Guides to more advanced levels of theory are also provided *Chemical Bonding [electronic Resource]* ,1991 **Chemical Bonds** ,2010 **The Nature of the Surface Chemical Bond** Thor N. Rhodin, Gerhard Ertl,1979 **Chemistry of Chemical Bonding** Jyoti Kumar Roshan,2008-01-01 **Molecules and the Chemical Bond** Henry A. Bent,2011 MOLECULES AND THE CHEMICAL BOND Chemistry Simplified This highly original book by a famous chemistry teacher about general chemistry in a new key may change how teachers teach Atomic Theory The Mole Concept and Avogadro s Constant The Gas Laws Solving Problems in Chemical Stoichiometry The Saturation and Directional Character of Chemical Affinity The Pauli Exclusion Principle Linnett s Double Spin Set Theory Pauling s Rules of Crystal Chemistry The Octet Rule Lewis Structures for O₂ NO CO SO₂ and SO₃ Construction of Bond Diagrams VSEPR Theory Dative Bonding Multicenter Bonding Bonding in Metals pH Calculations The Periodic Table The Energy Function and the First Law of Thermodynamics The Entropy Function and the Second Law of

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