

## Chemistry Chapter 8 Test

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- \_\_\_ 1. What type of bond would you expect in a compound of oxygen and sodium?  
 a. ionic  
 b. metallic  
 c. covalent  
 d. polar covalent
- \_\_\_ 2. The symbols  $\delta^-$  and  $\delta^+$  indicate which of the following?  
 a. electronegativity  
 b. electrostatic force  
 c. charge  
 d. partial charge
- \_\_\_ 3. Which of the following represents a nonpolar covalent molecule?  
 a. NaCl  
 b. Br<sub>2</sub>  
 c. HCN  
 d. Au
- \_\_\_ 4. The measure of an atom's ability to attract electrons to itself in a chemical bond is called \_\_\_\_\_.  
 a. electron affinity  
 b. polarity  
 c. electronegativity  
 d. ionization energy
- \_\_\_ 5. The polarity of a bond depends on the difference in the atoms' \_\_\_\_\_.  
 a. electron affinities  
 b. electronegativities  
 c. atomic radii  
 d. ionization energies
- \_\_\_ 6. What type of bond would you expect to form between nitrogen and potassium?  
 a. ionic  
 b. covalent  
 c. polar covalent  
 d. metallic
- \_\_\_ \* High electronegativity differences are found in \_\_\_\_\_ bonds.  
 a. metallic  
 b. ionic metal - nonmetal  
 c. covalent  
 d. Pauling
- \_\_\_ \* To achieve an octet when combining with potassium, would an atom of a Group 16 element be expected to lose, gain, or share electrons?  
 a. lose  
 b. gain  
 c. share  
 d. This cannot be determined from the given information.
- \_\_\_ 9. Which of the following does not exist as a diatomic molecule?  
 a. argon  
 b. hydrogen  
 c. chlorine  
 d. oxygen  
 H N O F Cl Br I  
 Gas: H<sub>2</sub> N<sub>2</sub> O<sub>2</sub> F<sub>2</sub> Cl<sub>2</sub> Br<sub>2</sub> I<sub>2</sub>
- \_\_\_ 10. Which of these is not a covalent molecule?  
 a. F<sub>2</sub>  
 b. CO  
 c. LiF  
 d. NH<sub>3</sub>
- \_\_\_ 11. A dash in a Lewis structure represents \_\_\_\_\_.  
 a. one electron  
 b. two electrons  
 c. four electrons  
 d. unpaired electrons
- \_\_\_ 12. The central atom in a Lewis structure is usually the atom \_\_\_\_\_.

# Chapter 5 Chemistry Test Answers

**S Ben Porath**



## **Chapter 5 Chemistry Test Answers:**

## **Chapter 5 Chemistry Test Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter 5 Chemistry Test Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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