

# 7.3

## BONDING IN METALS

### Section Review

#### Objectives

- 1. Describe metallic bonding
- 2. Describe the delocalized electron model
- 3. Explain the properties of metals

#### Vocabulary

- metallic bond
- metal

#### Part A Completion

Use the following words to complete each sentence. Write the word in the space provided in the answer column. Each word can be used once, more than once, or not at all.

- delocalized electron model    1.    metal conductivity
2.    The delocalized electron model is used to explain the metallic bond.
3.    The delocalized electron model explains the properties of metals.
4.    The delocalized electron model explains the properties of metals.
5.    The delocalized electron model explains the properties of metals.
6.    The delocalized electron model explains the properties of metals.
7.    The delocalized electron model explains the properties of metals.
8.    The delocalized electron model explains the properties of metals.

- 9.    The delocalized electron model explains the properties of metals.
- 10.    The delocalized electron model explains the properties of metals.
- 11.    The delocalized electron model explains the properties of metals.
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- 18.    The delocalized electron model explains the properties of metals.
- 19.    The delocalized electron model explains the properties of metals.
- 20.    The delocalized electron model explains the properties of metals.

#### Part B True/False

Circle whether each statement is true (T) or false (F).

1.    The delocalized electron model explains the properties of metals.    T
2.    The delocalized electron model explains the properties of metals.    F

# Bonding In Metals Section Review Answers Key

**J Rink**



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