

# Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

Chemistry

Physical Science

## SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a homogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution.

1. Define the term solvent and provide an example.

2. Define solubility, and how is it measured?

3. Distinguish between concentrated and dilute.

4. List the properties of bases and give an example.

5. Explain the significance of the pH scale.

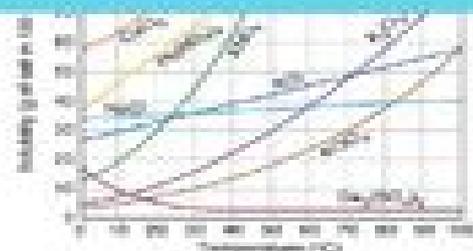
6. How does the concentration of an acid affect its pH?

7. Explain the relationship between pH and pOH.

8. Explain the difference between strong acids and weak acids.

9. Explain the difference between strong bases and weak bases.

10. Explain the difference between a strong acid and a weak acid.



11. Describe how you would use the graph to determine the amount of  $\text{NaCl}$  that can be dissolved in 100 grams of water at  $60^\circ\text{C}$ .

12. Which salt has the greatest solubility at  $20^\circ\text{C}$ ?



# ANSWER KEY INCLUDED

# Acids And Bases Study Guide Key

**Ensheng Dong**



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